

Transforming Urban Landscapes with Responsive Materials

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TEXAS A&M
UNIVERSITY.



Canadian Centre canadien
Light de rayonnement
Source synchrotron

Energy & Buildings



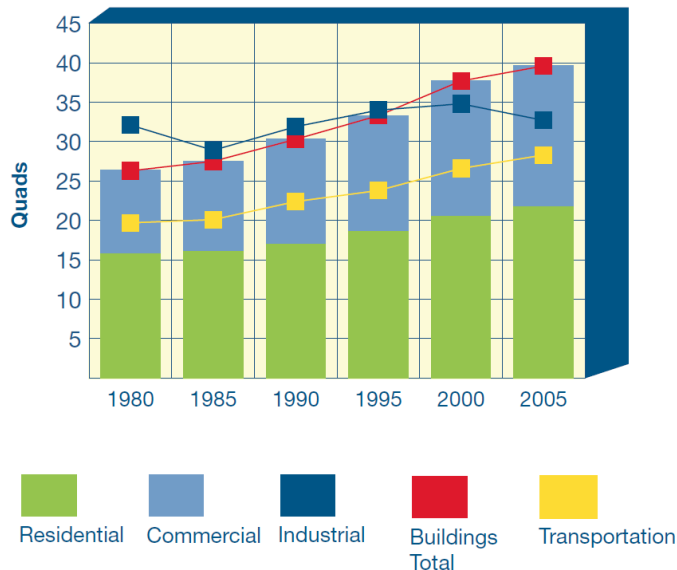
- We are in the midst of one of the greatest migrations in the history of mankind
- >50% of the world's population lives in urban areas
- 300M to move to cities in China alone
- 2B m² of new construction each year in China, >80% energy inefficient



What do Buildings Cost?

Buildings account for 72 percent of U.S. electricity use and 36 percent of natural gas use

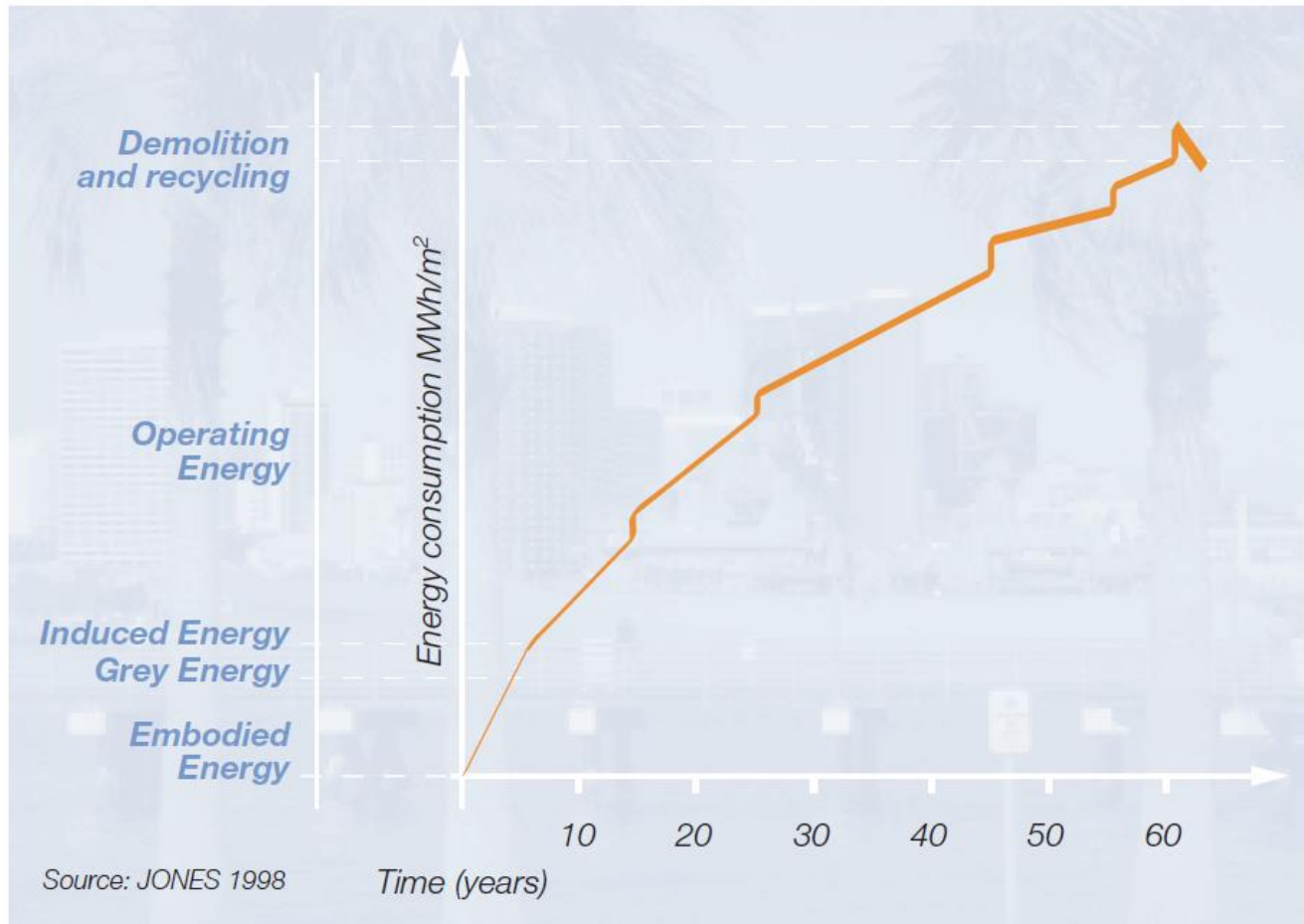
Figure 8
Growth in Buildings Energy Use Relative to Other Sectors



Buildings account for 40 percent of all energy use in the United States. This sector consumes more energy than either industrial or transportation, surpassing industrial as the number one consuming sector in 1998. Both residential and commercial building energy use are growing, and represent an ever-increasing share of U.S. energy consumption. While residential energy consumption exceeds commercial, the latter has been increasing more rapidly, rising from just 14 percent of total U.S. energy consumption in 1980 to 18 percent by 2005, a 70 percent increase.

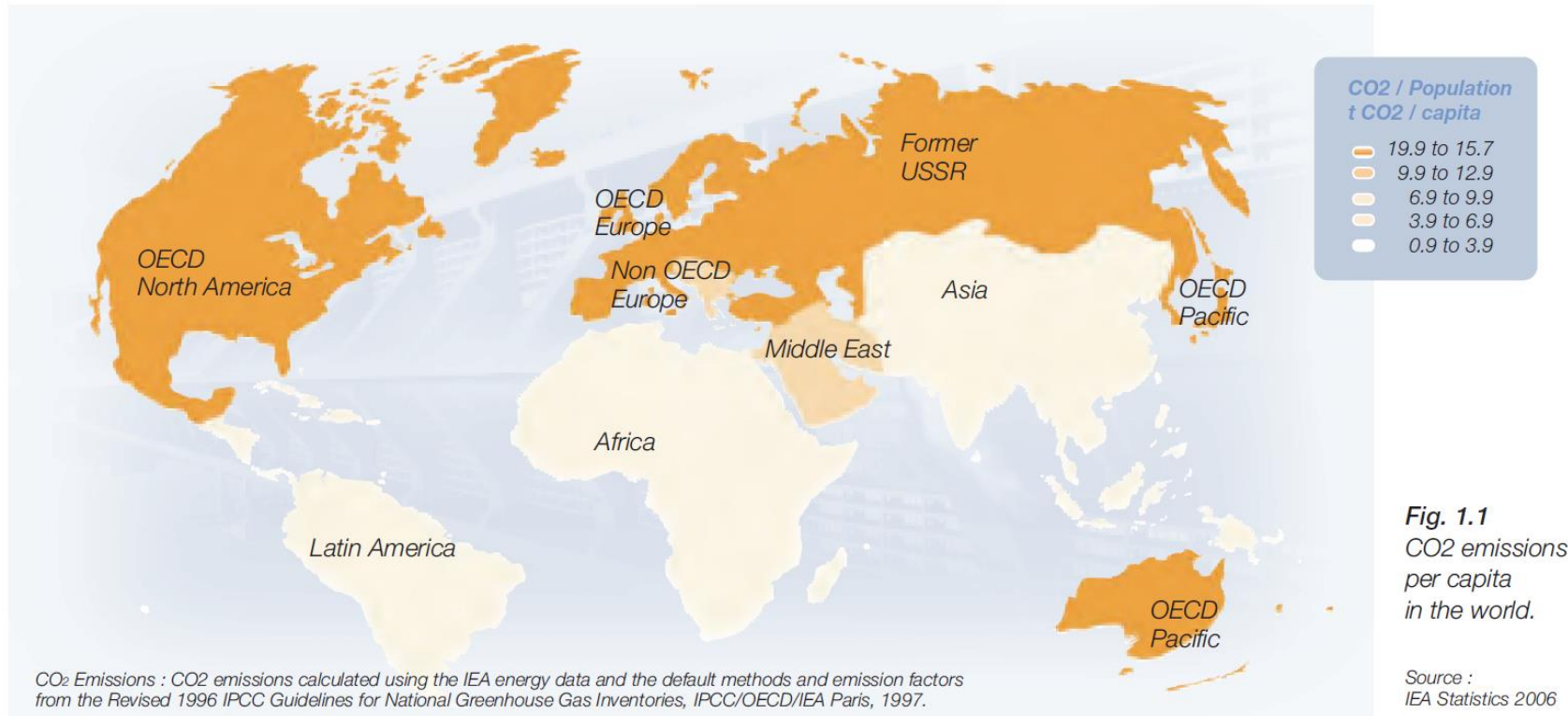
- 30-40% of energy usage worldwide is from buildings
- Equivalent to 2500 megatons oil equivalent

What do Buildings Really Cost?



- Choices we make now will benefit/haunt us in the decades to come

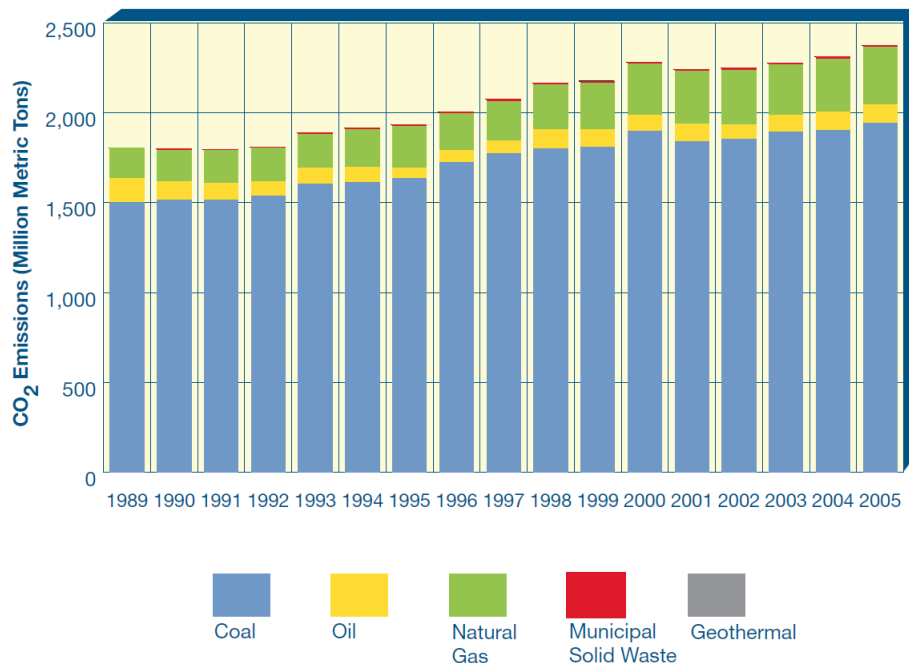
The Real Costs: CO₂ Emissions



Going up in smoke...

U.S. buildings currently contribute 9 percent of the world's carbon dioxide emissions

Figure 12
Contributors to Electricity CO₂ Emissions



The growth in buildings energy consumption has resulted in carbon dioxide emissions rising from about a third of total U.S. emissions in 1980 to almost 40 percent by 2005. This is a function of the increase in buildings electricity use, 70 percent of which is dependent on fossil fuels. Despite recent efforts to use cleaner coal technologies, the majority of carbon dioxide emissions are still attributable to coal. Both geothermal and municipal solid waste represented negligible amounts of carbon dioxide emissions: 0.4 and 11 million metric tons in 2005, respectively.

Where does this energy come from? China as a case study

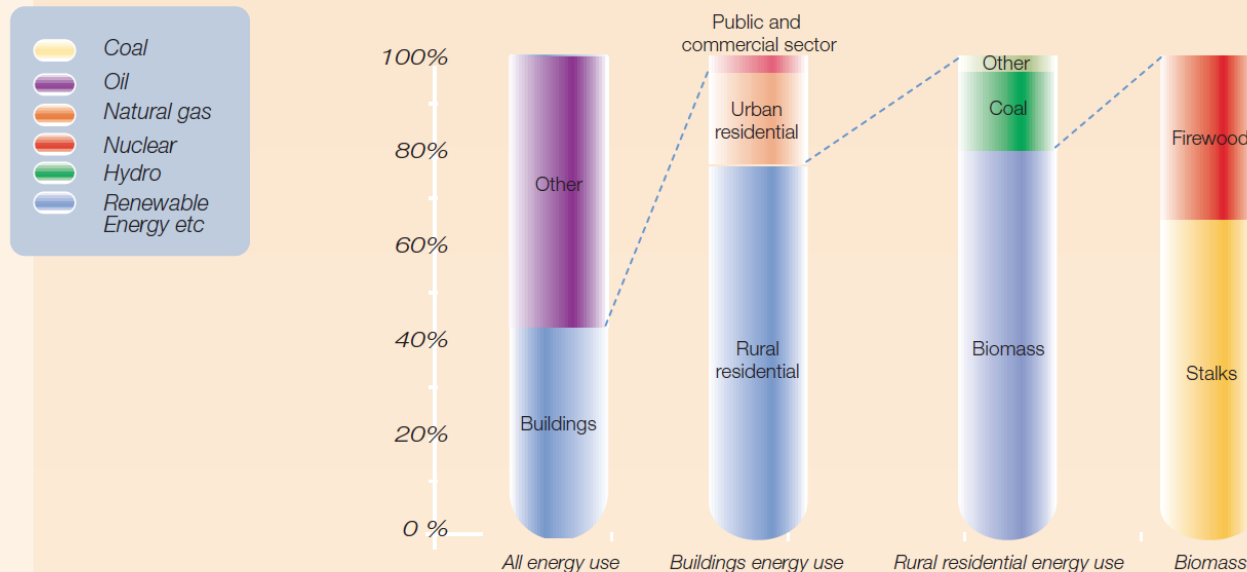


Fig. 2.11
Distribution of Chinese energy use.

Source : Yutaka et al. 2005.



Where does the Energy Go?

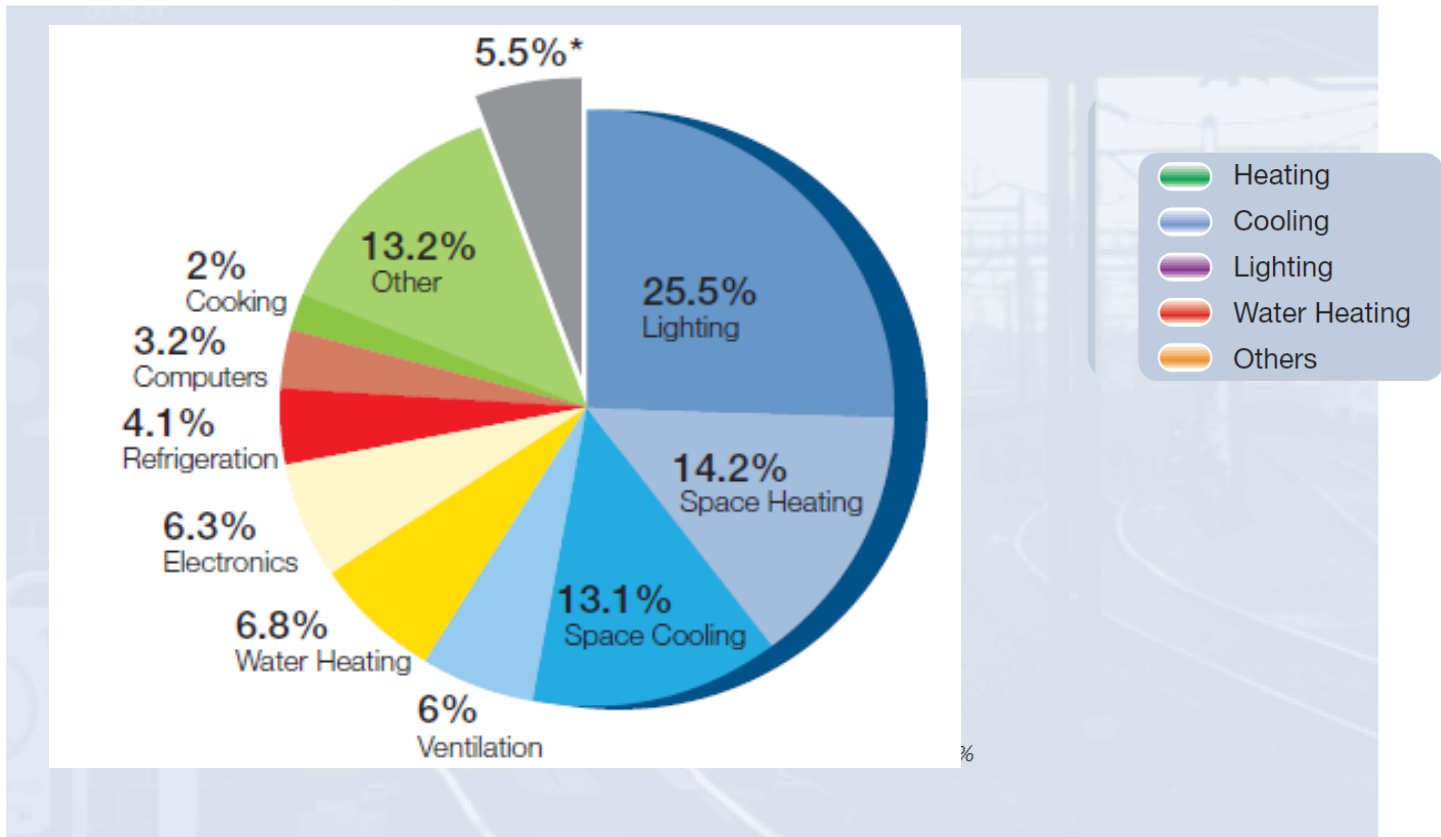


Fig. 2.14

Shares of different energy end-use purposes for residential and commercial buildings in some countries.

Source : Al-Sayed Omar Assem and Al-Ragom 2005, CMIE 2001, Sustainable Energy Authority Victoria 2004, U.S. Department of Energy 2006, Office of Energy Efficiency; Natural Resources Canada 2006.



- 40% of energy consumed in Mumbai, India goes toward space cooling
- China added 50M home air-conditioning units in 2010
- By 2050, 27% of all global warming will be due to coolant gases

©NYTimes



A Deeper Dive...

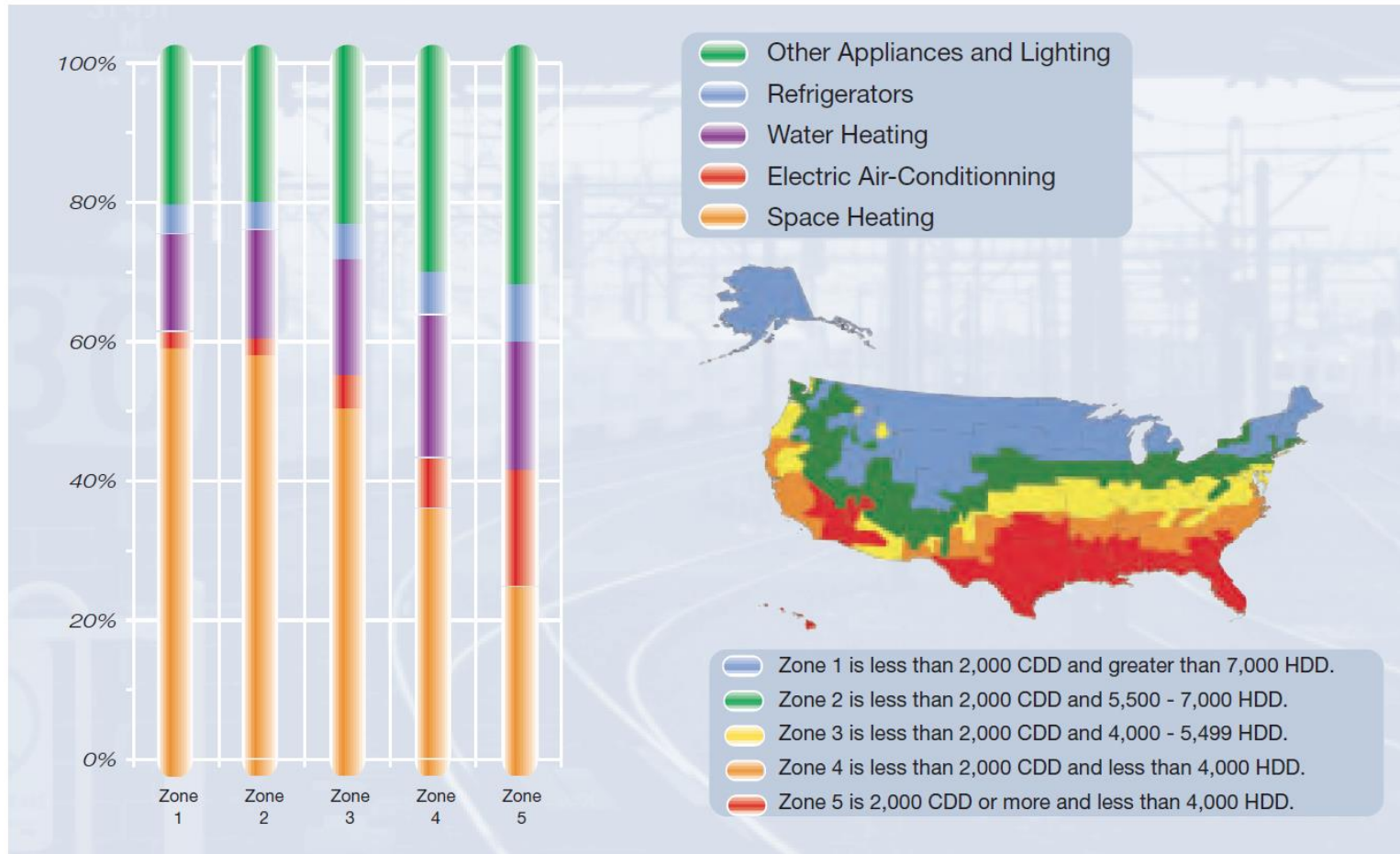
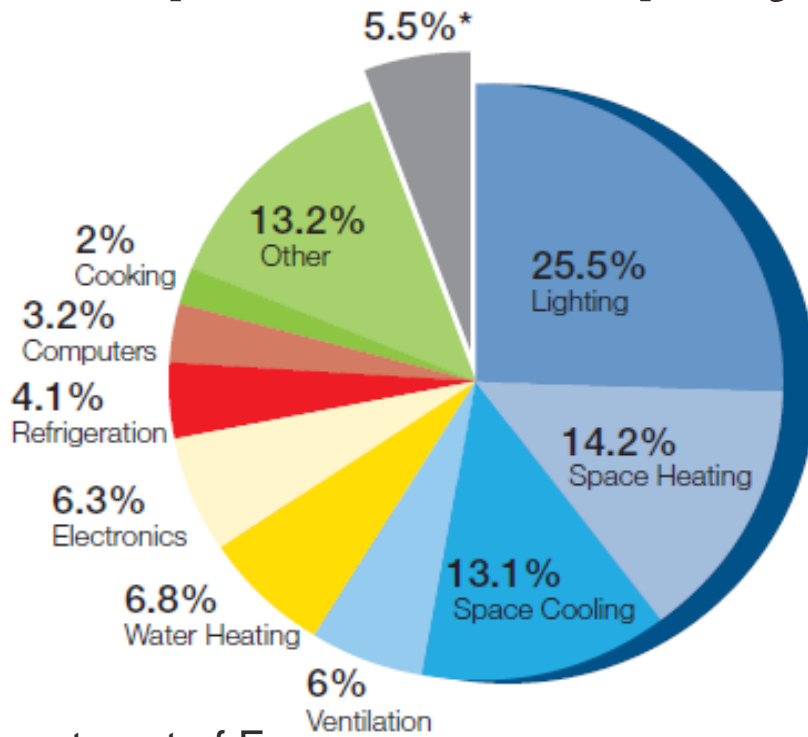


Fig. 2.15
Energy consumption in US residential buildings.

Source : US EIA 2001



The price we pay for our buildings



Department of Energy



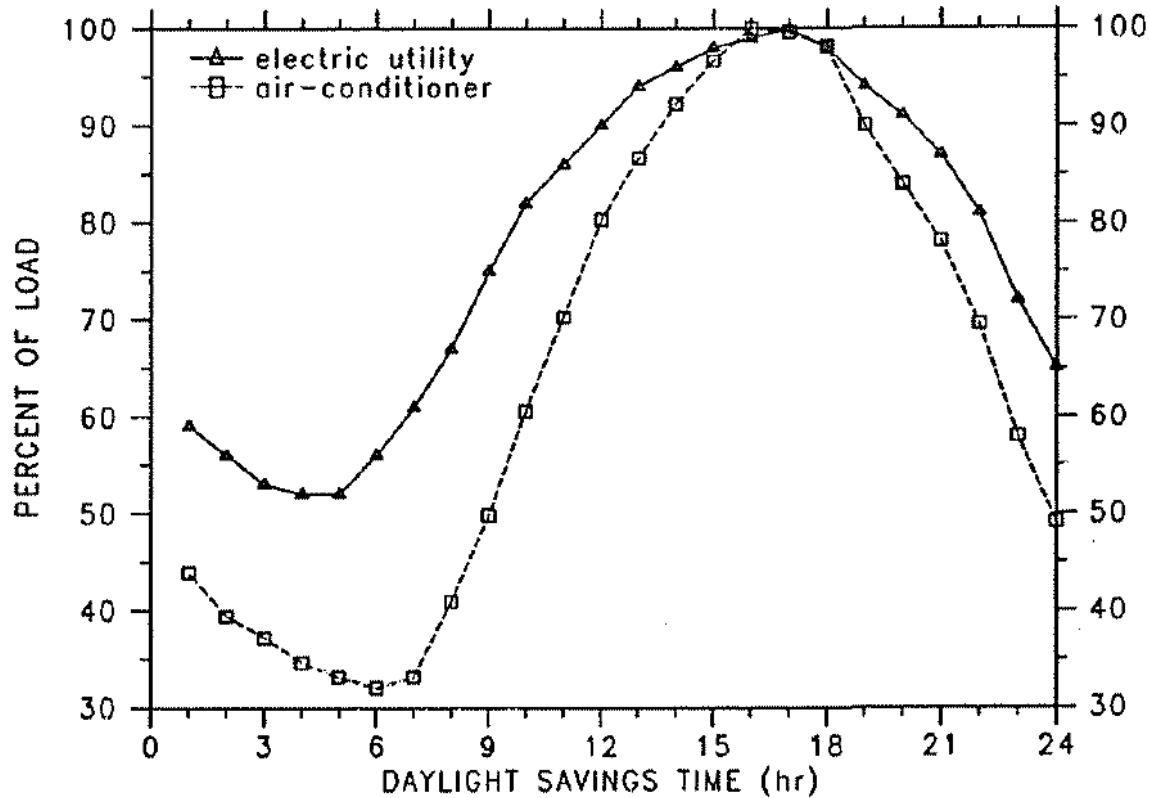
- ❑ \$15B
- ❑ 140M metric tons of CO₂



US Department of Energy, Energy Efficiency Trends in Residential and Commercial Buildings, Washington DC 2008



We live in the grid...



Average peak-summer-day load shape for Florida electric utility and average load shape of 58 residential central air conditioners.

Conditioned Environments



- 7 Billion Gallons of Gasoline
- 50 Million Metric Tons of CO₂
- Refrigerant Leakage Equivalent to 50 Million Tons of CO₂

The Building Envelope



The “building envelope” refers to the external walls, windows, roof, and floor of a building. This barrier between indoors and outdoors is important with regards to ventilation and insulation of a conditioned space.

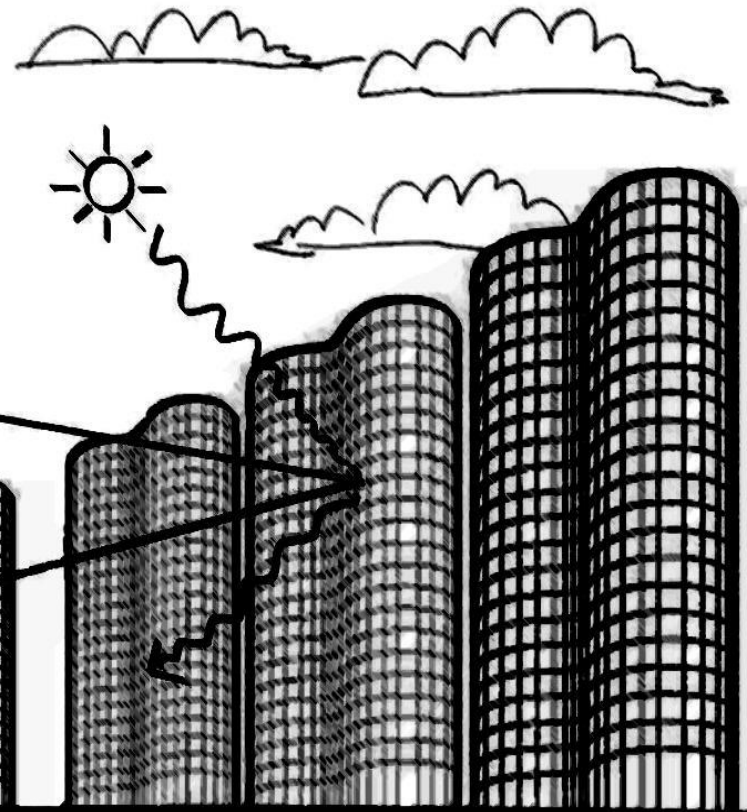
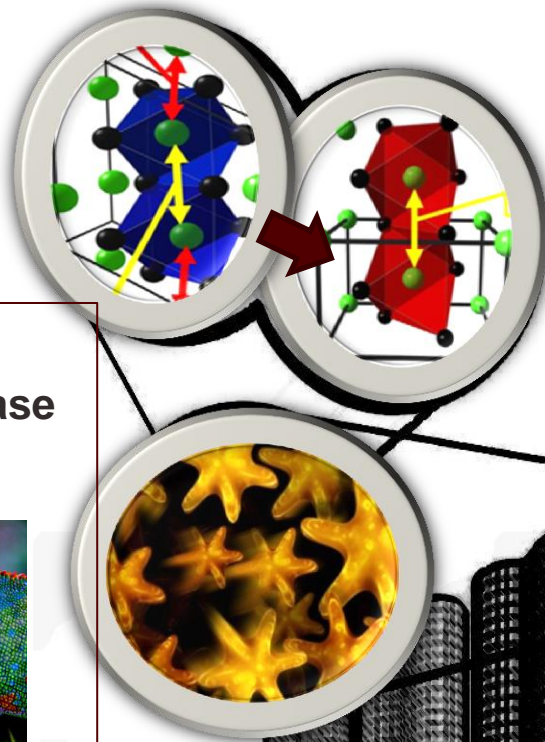
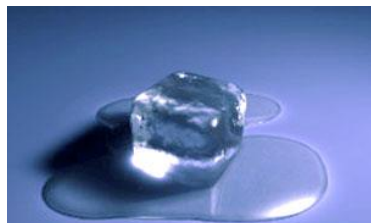
What does this have to do with physics?



Buildings that change with the climate...

-adaptable to specific climates and to night/day, summer/winter

- Physics and chemistry of phase transformations



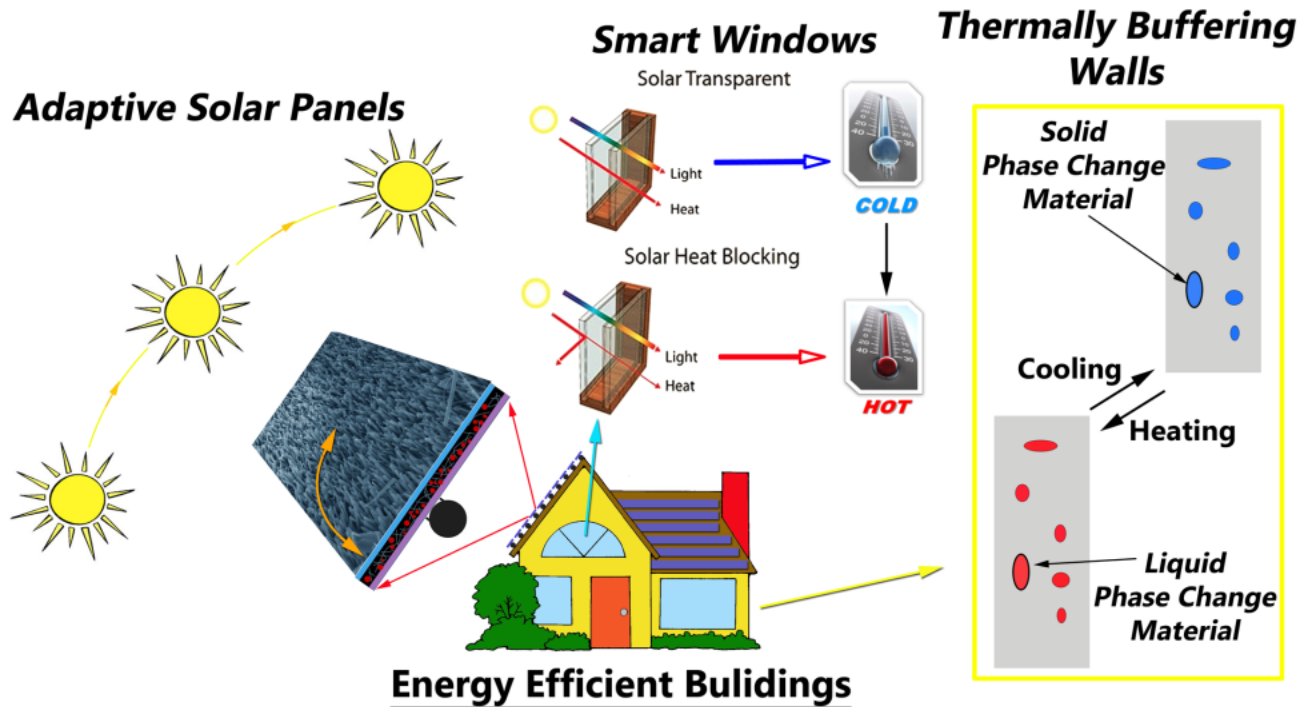
Zero emission buildings that sense and respond to the climate

Can we endow our windows and walls with intelligence?



Smart Windows" seen at light and dark settings. | Photo Courtesy of SAGE Electrochromics, Inc., by Susan Fleck Photography

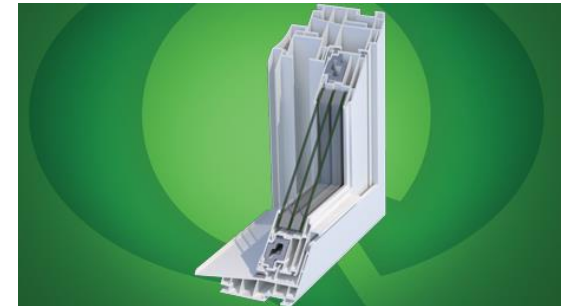
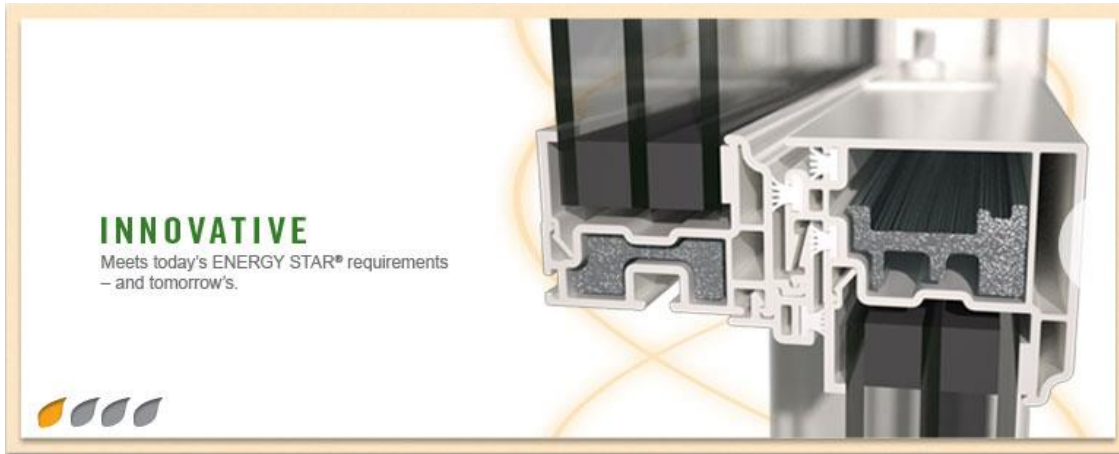
Buildings of the Future: Living and Breathing Constructs



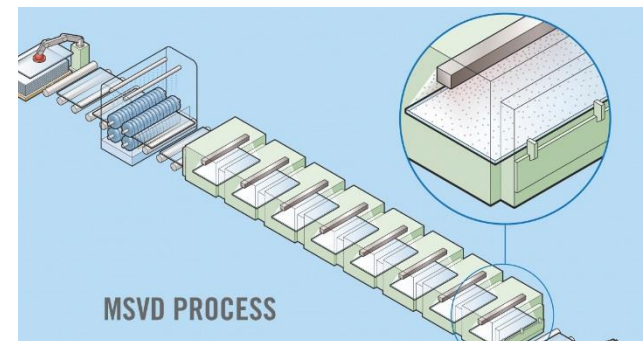
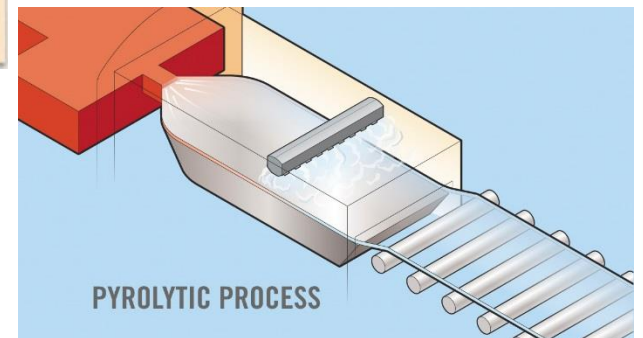
- Local energy conservation
- Distributed energy storage



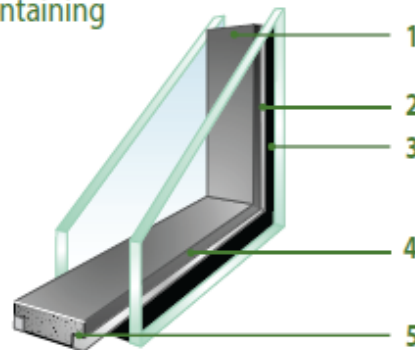
What's Inside your Windows?



Low-E Glass

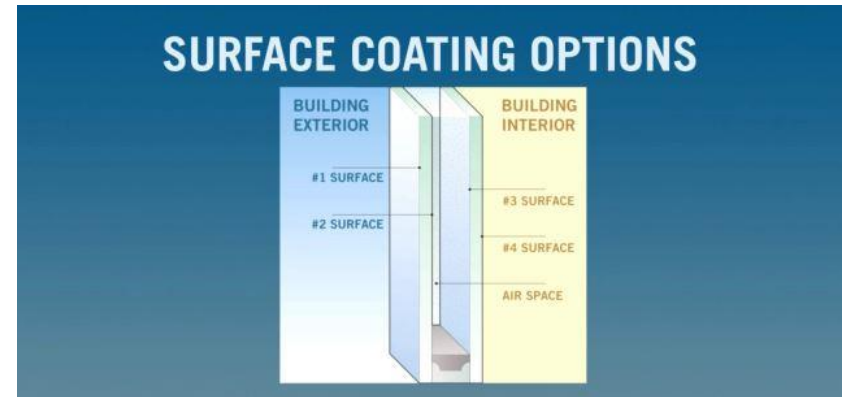
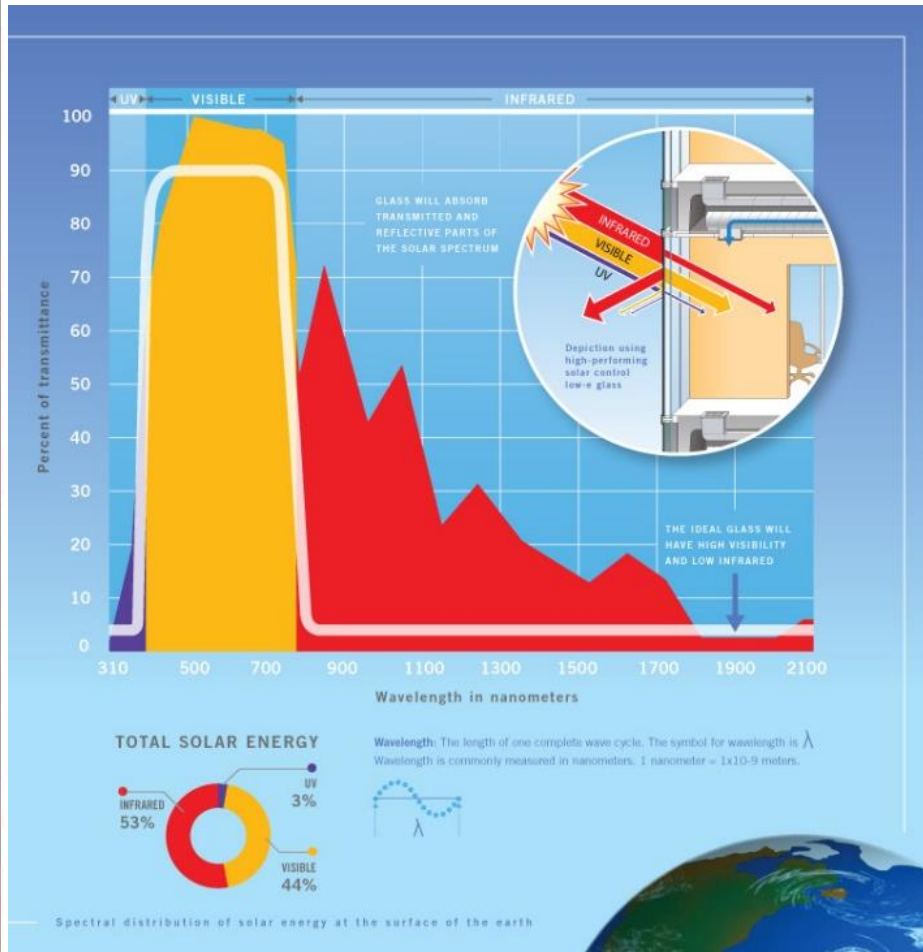


1. Thermoset structural silicone foam containing NO-Metal with integral 3A desiccant
2. PIB primary seal
3. Silicone, Polyurethane, Polysulfide, DSE/DSA's or Hot Melt secondary seal
4. Pressure-sensitive acrylic adhesive
5. Pre-applied advanced multi-layer vapor barrier



A lot of physics and chemistry!

State-of-the Art: Low-E-Glass



**Effective at Reflecting Long-Wavelength IR: “Thermos” Principle
Static - Silver! - High-Vacuum Processing**

Measuring up a Window

U-Value

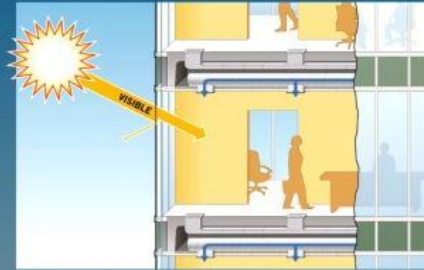


- **U-Value** is the rating given to a window based on how much heat loss it allows.

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Visible Light Transmittance

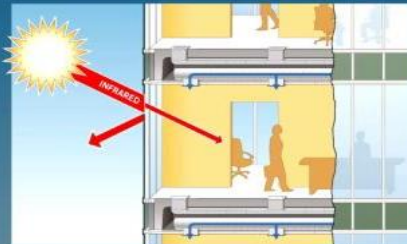


- **Visible Light Transmittance** is a measure of how much light passes through a window.

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Solar Heat Gain Coefficient

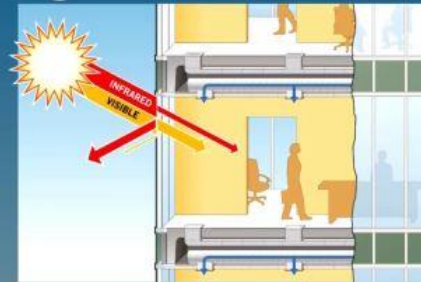


- **Solar Heat Gain Coefficient** is the fraction of incident solar radiation admitted through a window, both directly transmitted and absorbed and subsequently released inward. The lower a window's solar heat gain coefficient, the less solar heat it transmits.

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Light to Solar Gain



- **Light to Solar Gain**, which is the ratio between the window's Solar Heat Gain Coefficient (SHGC) and its visible light transmittance (VLT) rating.

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Light versus Heat

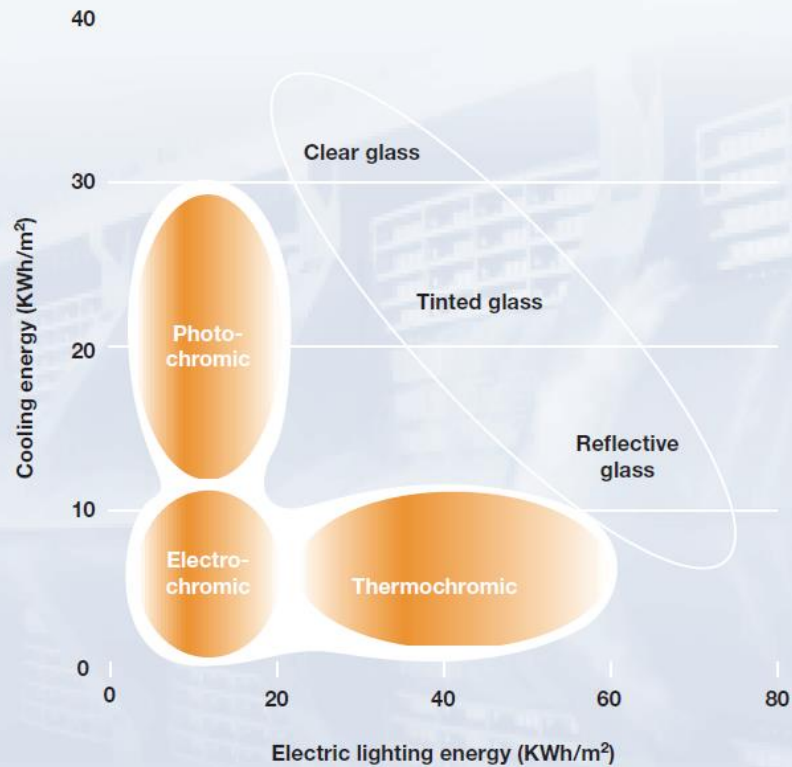


Figure 3.2
Lighting energy versus cooling energy for different glazing types.



Adaptive Materials

Respond in a reversible manner to changes in temperature, humidity, or light

-accompanying changes in optical, electronic, or thermal properties allow for modulation of energy consumption within a building

Materials development
Building integration
Efficiency and RoI metrics
Durability
Aesthetic acceptability



Responsive Materials for Smart Windows

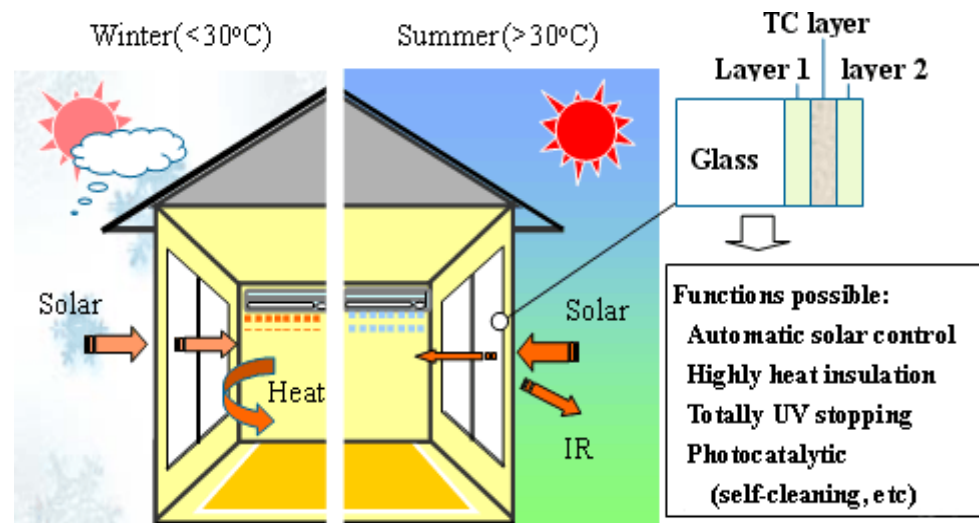
Phase Change Materials: Two or more optical states that can be switched in response to temperature, light, voltage, or humidity

Electronic Crystals: large switching of carrier density induced as a result of electron—phonon or electron—electron interactions

Electrochromics: Changes in redox states or polaronic confinement (inter-valence charge transfer bands)

Phase-Change Materials: Melting or crystalline—amorphous transitions

Lyotropic/Thermotropic Transitions: Phase segregation of change in conformation of molecular chains

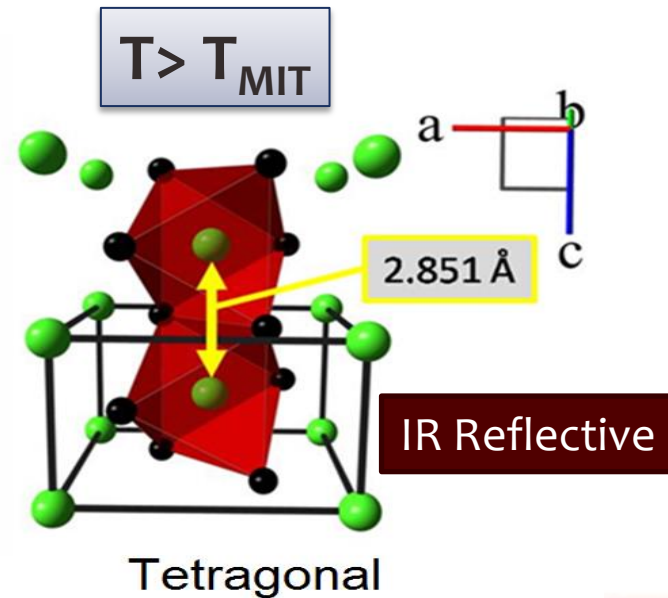
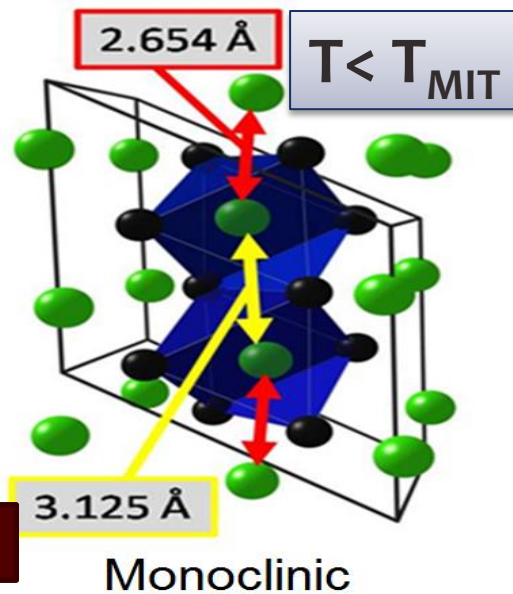
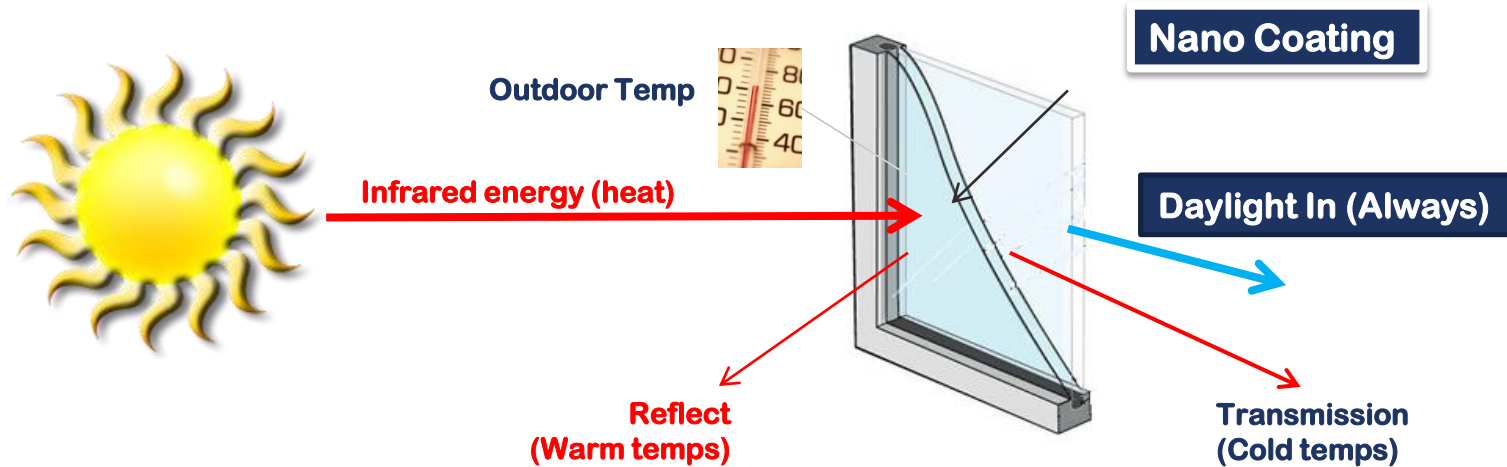


Thermochromic
Electrochromic
Photochromic
Gasochromic

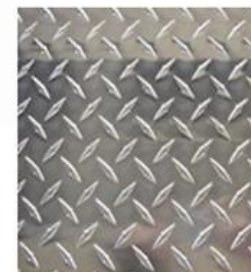
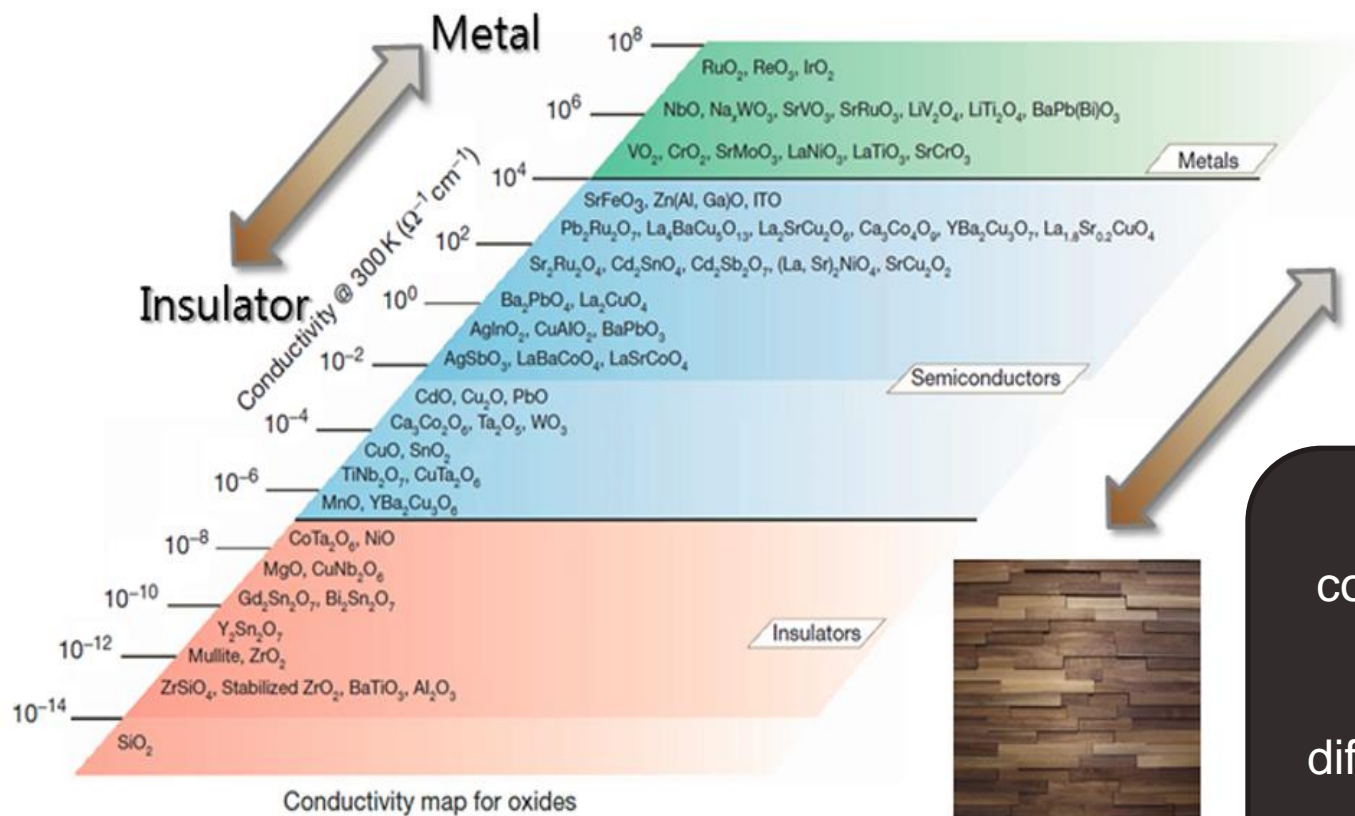
Image from AIST Japan



Dynamic Windows: Light without Heat?



Electronic Phase Transitions



Electrical conductivity spans 23 orders of magnitude differences at room temperature

P. M. Marley, G. A. Horrocks, K. E. Pelcher, and S. Banerjee,* Transformers: The Changing Phases of Low-Dimensional Vanadium Oxide Bronzes, Chemical Communications 2015, DOI: 10.1039/C4CC08673B.

P. Edwards, V. Kuznetsov, D. Slocombe and R. Vijayaraghavan, Comprehensive Inorganic Chemistry II, 2013



Single-Particle vs. Many-Body Insulators

Insulators due to electron-ion interaction (**single-particle physics**):

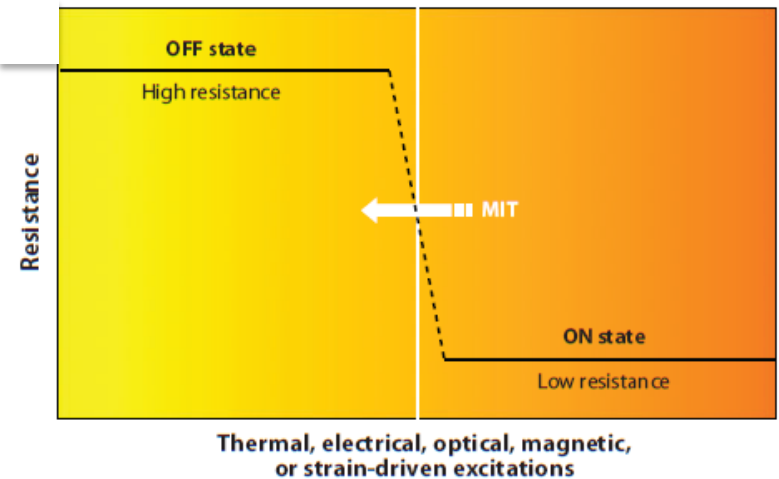
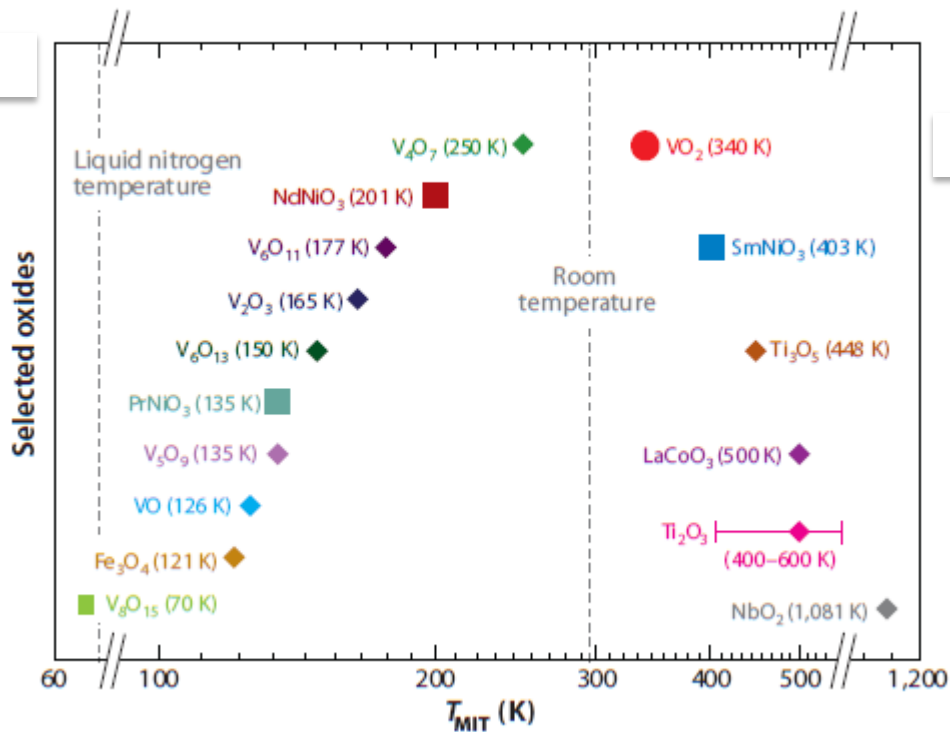
- Band Insulators (electron interacts with a periodic potential of the ions → **gap in the single particle spectrum**)
- Peierls Insulators (electron interacts with static lattice deformations → **gap**)
- Anderson Insulators (electron interacts with the disorder—such as impurities and lattice imperfections)

Mott Insulators due to electron-electron interaction (many-body physics leads to the **gap in the charge excitation spectrum**):

- Mott-Heisenberg (antiferromagnetic order of the pre-formed local magnetic moments below Néel temperature)
- Mott-Hubbard (no long-range order of local magnetic moments)
- Mott-Anderson (disorder + correlations)
- Wigner Crystal (**Coulomb interaction dominates at low density of charge**, $r_s(2D) = E_{e-e}/E_F = n_s^{-1/2}/n_s = 33$ or $r_s(3D) = 67$, thereby localizing electrons into a **Wigner lattice**)

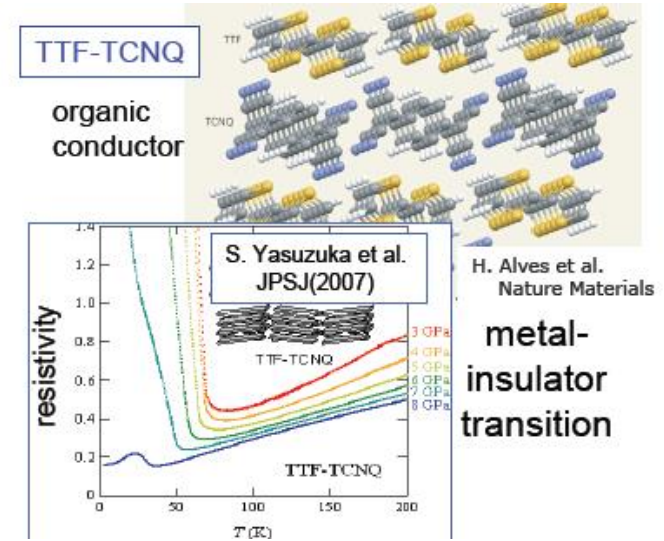
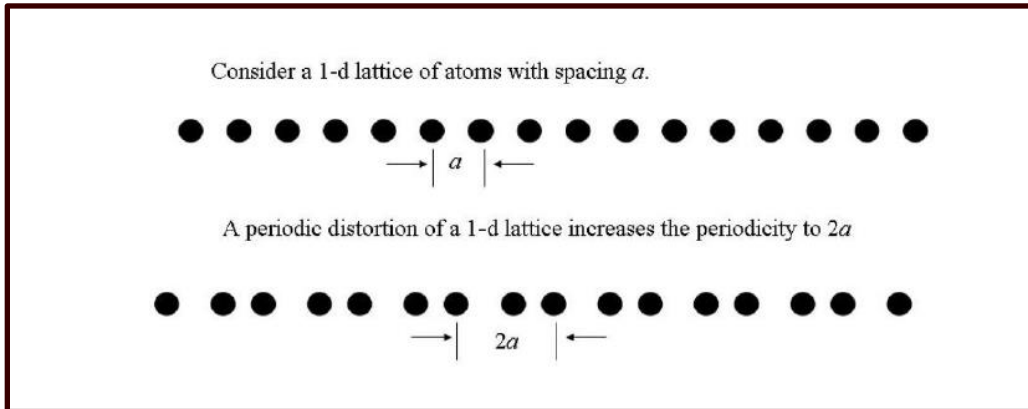
Compounds that should be metals but aren't!

Metal—Insulator Transitions

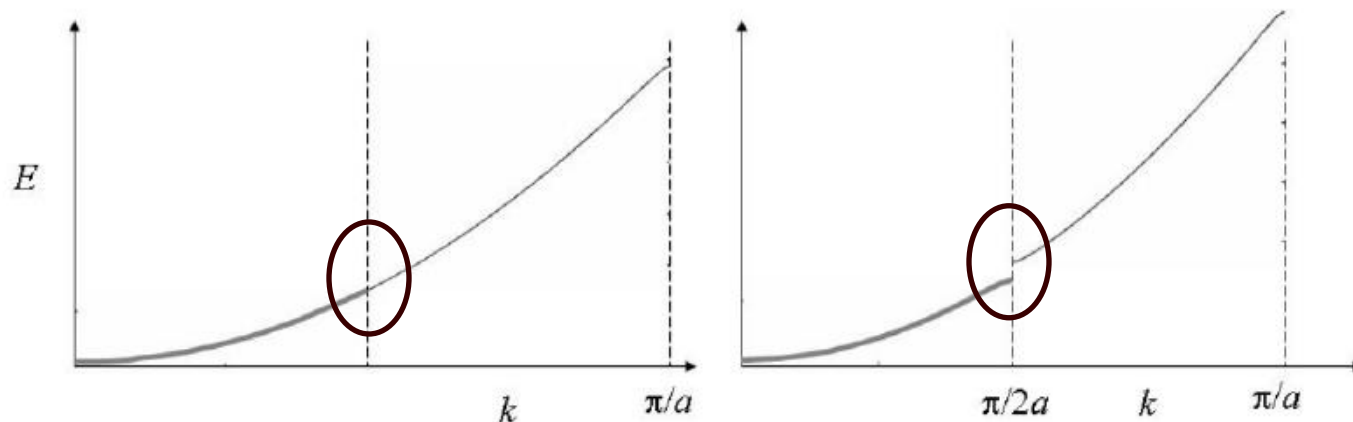


Design of strong electron correlated materials remains one of the “grand challenges” of science

Electron—Phonon Insulator: Peierl's Insulator

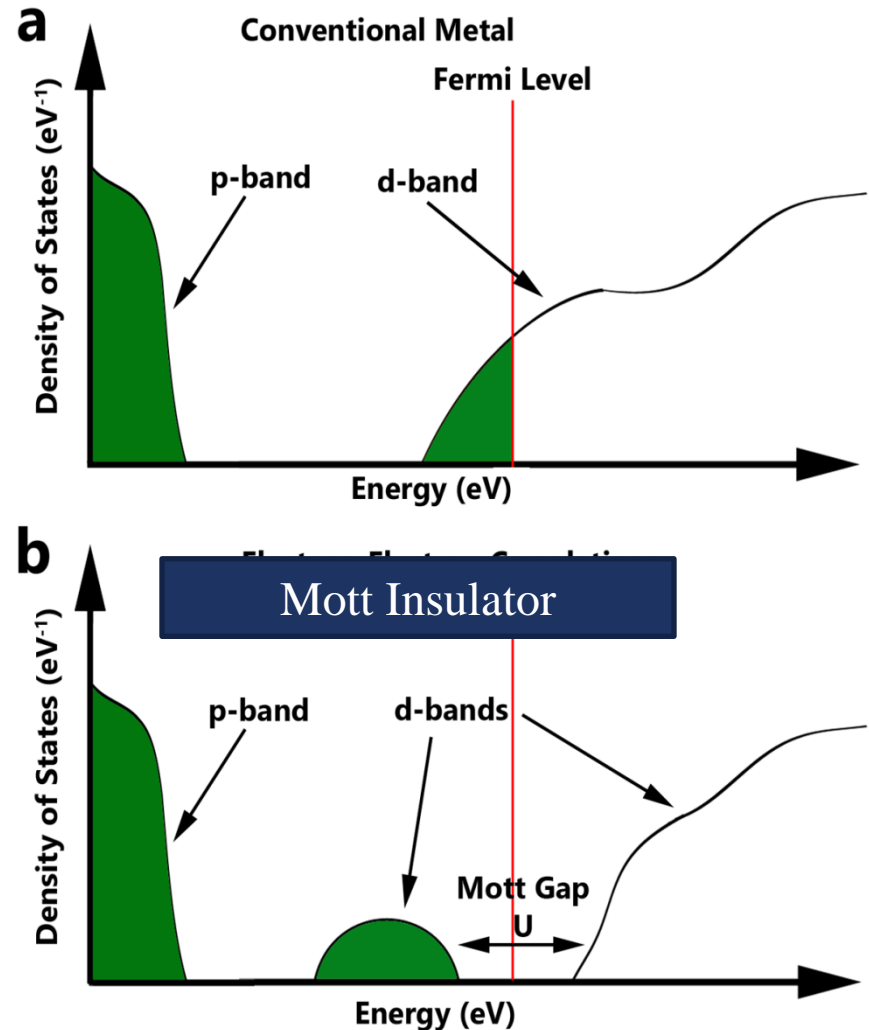


A gap appears at the end of the filled states: filled states are lowered in energy and unfilled states are raised in energy



Mott's Initial Idea for Correlation-Induced Gap

- Two electrons localized at a particular site of distance (d) will repel each other with an energy, U (Mott-Hubbard correlation energy).
- The resulting band is split into Lower and Upper Hubbard Bands (LHB and UHB, respectively) creating a band gap.



The Mott Metal-Insulator Transition

W = electronic band width

At a critical ratio of $W:U$ the bands overlap and metallic behavior sets in.

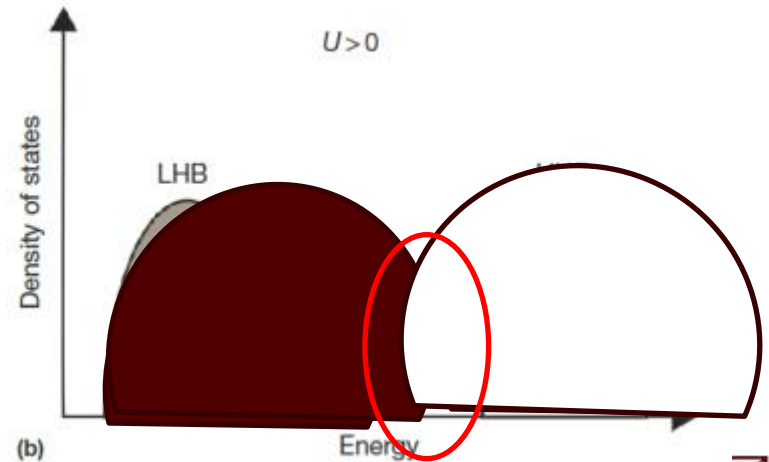
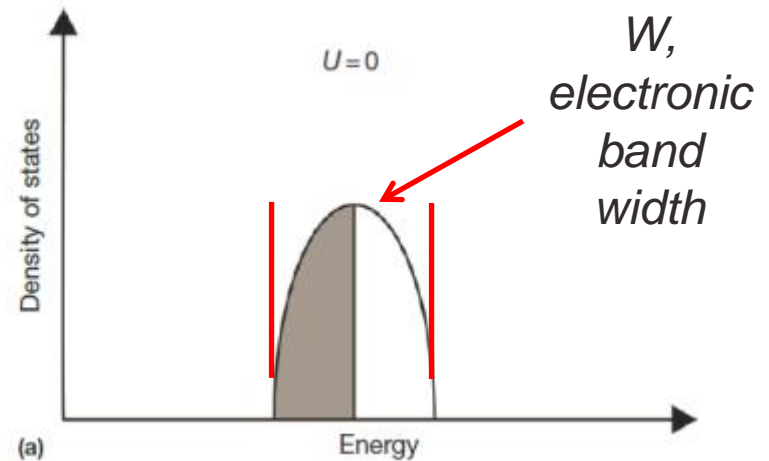
Mott criterion for MIT:

$$n_c^{1/3} a_h^* = 0.25$$

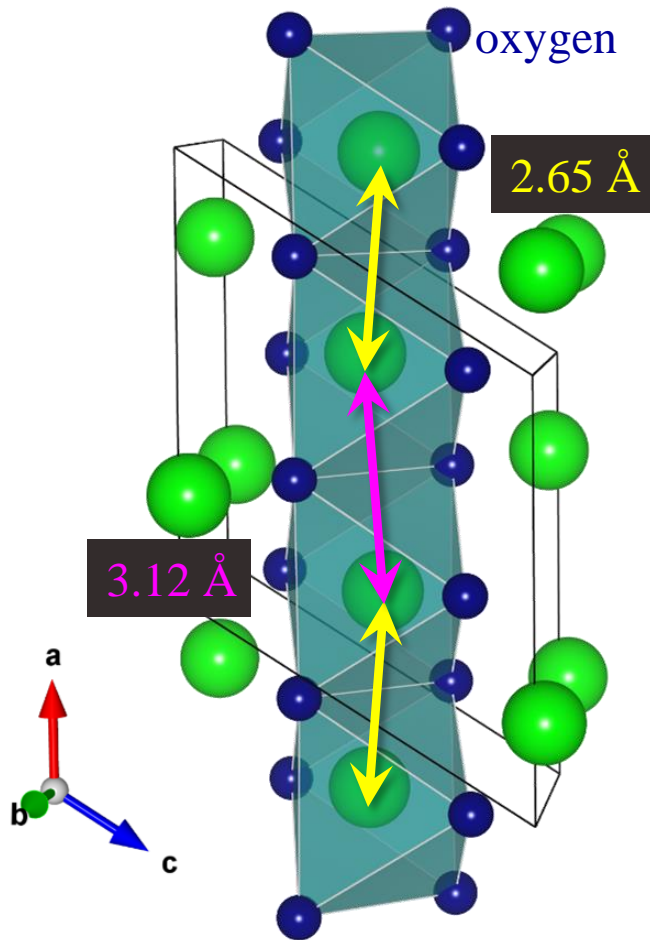
n_c = critical carrier concentration

a_h^* = Bohr radius of the localized dopant

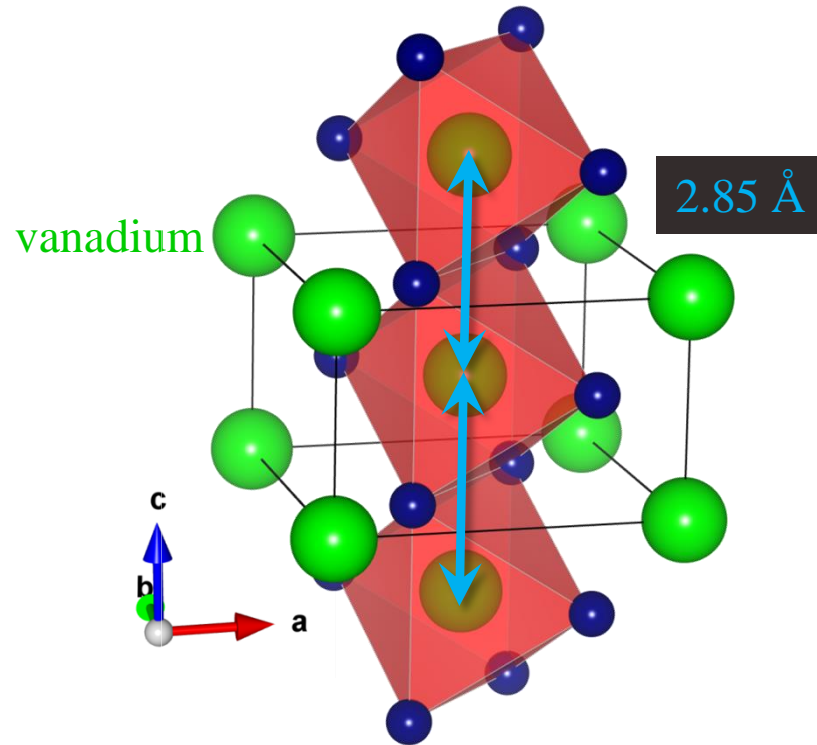
Since the critical distance between dopant centers (d_c) is $\sim n_c^{-1/3}$, the Mott MIT will occur when $d_c \sim 4a_h^*$. i.e. when the distance between dopant centers is about four times the Bohr radius of the localized dopant.



Vanadium Oxide



Low Temperature Monoclinic Phase



High Temperature Rutile Phase

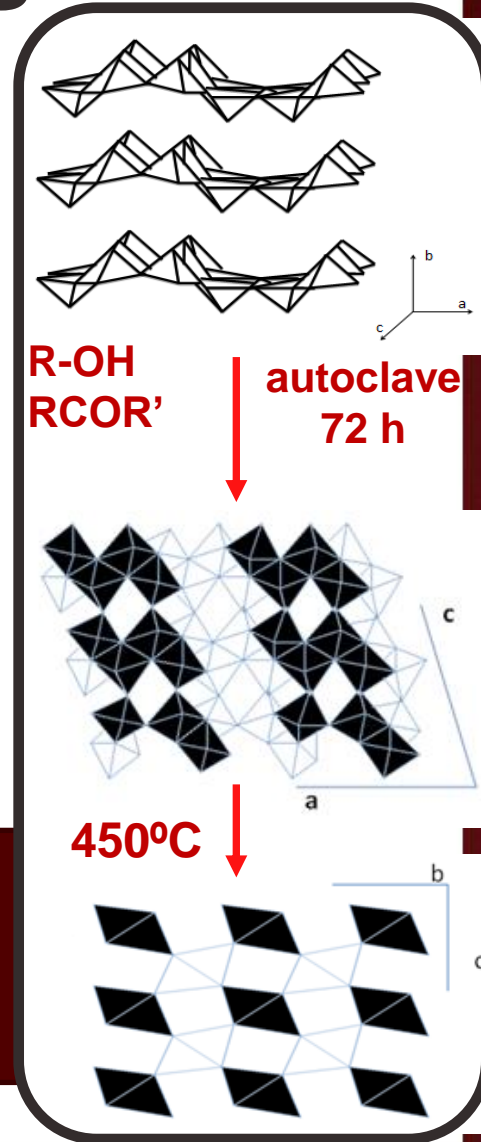
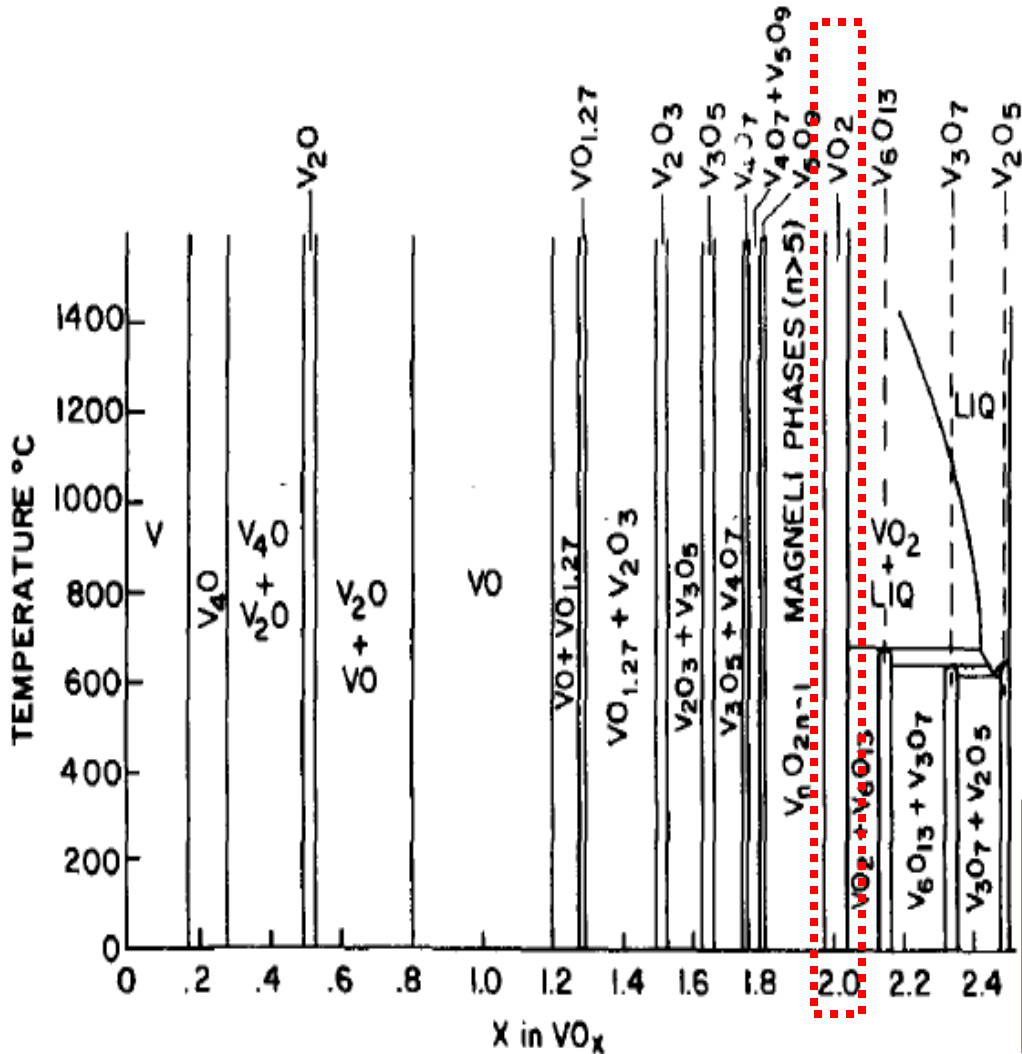
Challenges with VO₂ Windows?

Bulk or thin film VO₂ cracks upon thermal cycling-can't withstand the thermal stresses upon repeated cycling

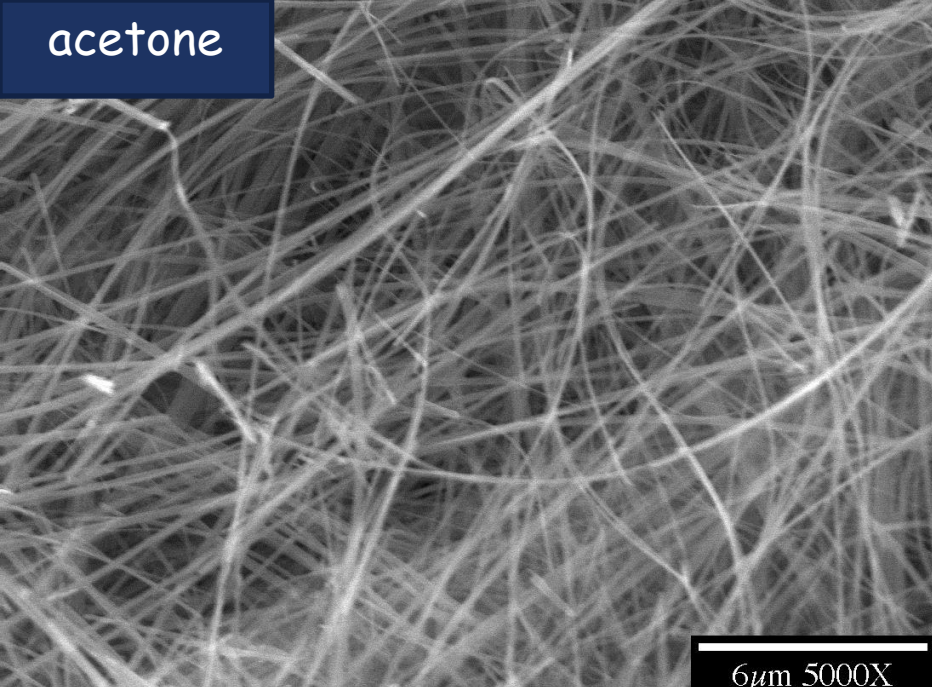
The intrinsic metal—insulator transition temperature of 67°C is too high to be useful in a practical window.

Difficult (impossible) to prepare VO₂ with careful control of stoichiometry in large amounts-most “high-quality” VO₂ confined to epitaxial thin films grown under ultra-high-vacuum conditions. No scalable methods to prepare powders

VO₂ as a Synthetic Target

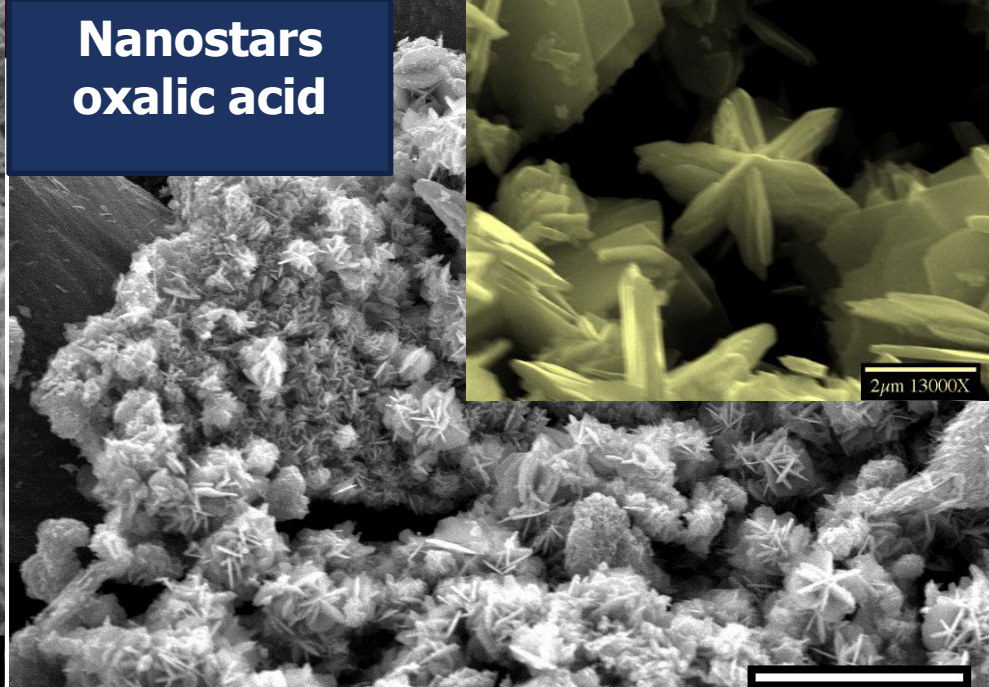


acetone



6 μm 5000X

Nanostars
oxalic acid

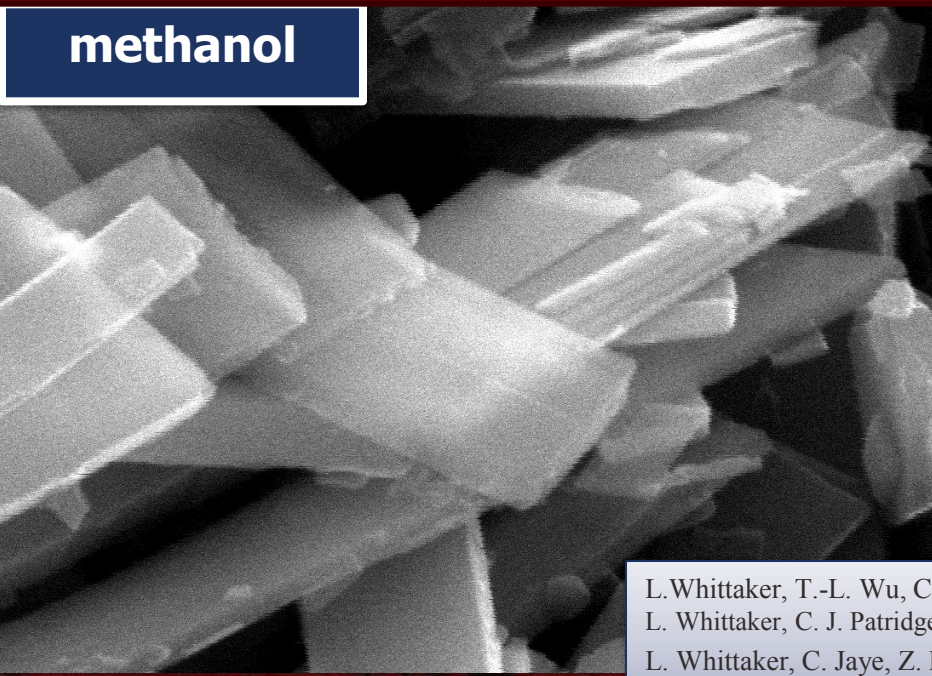


20 μm 1300X

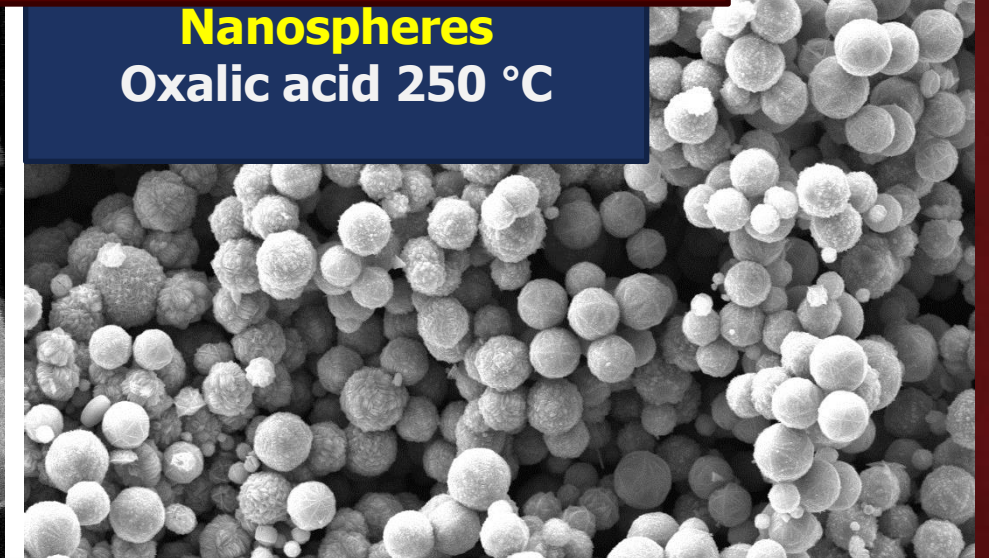
2 μm 13000X

Nano and microstructures can be cycled thousands of times without fracture

methanol



Nanospheres
Oxalic acid 250 °C

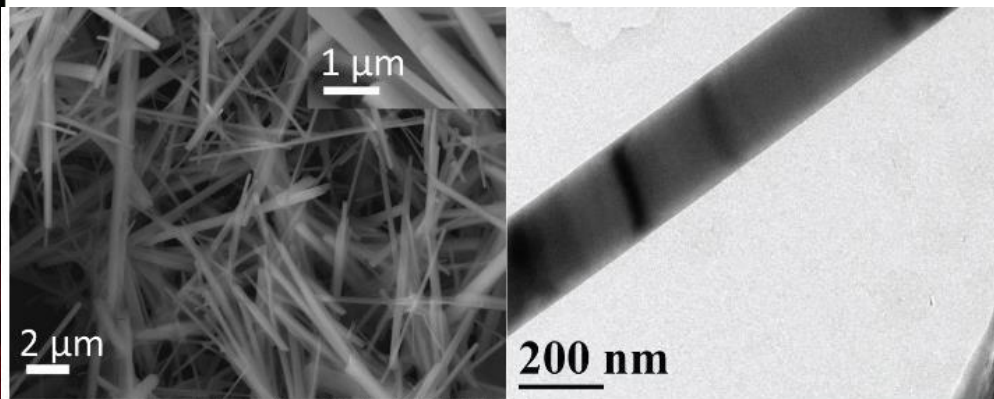
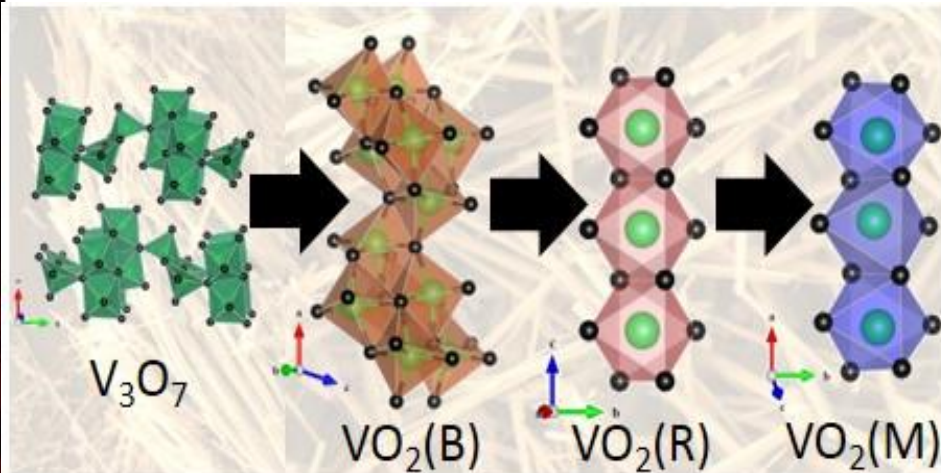


L. Whittaker, T.-L. Wu, C. J. Patridge, G. Sambandamurthy, and S. Banerjee, *J. Mater. Chem.* **2011**, *21*, 5580.

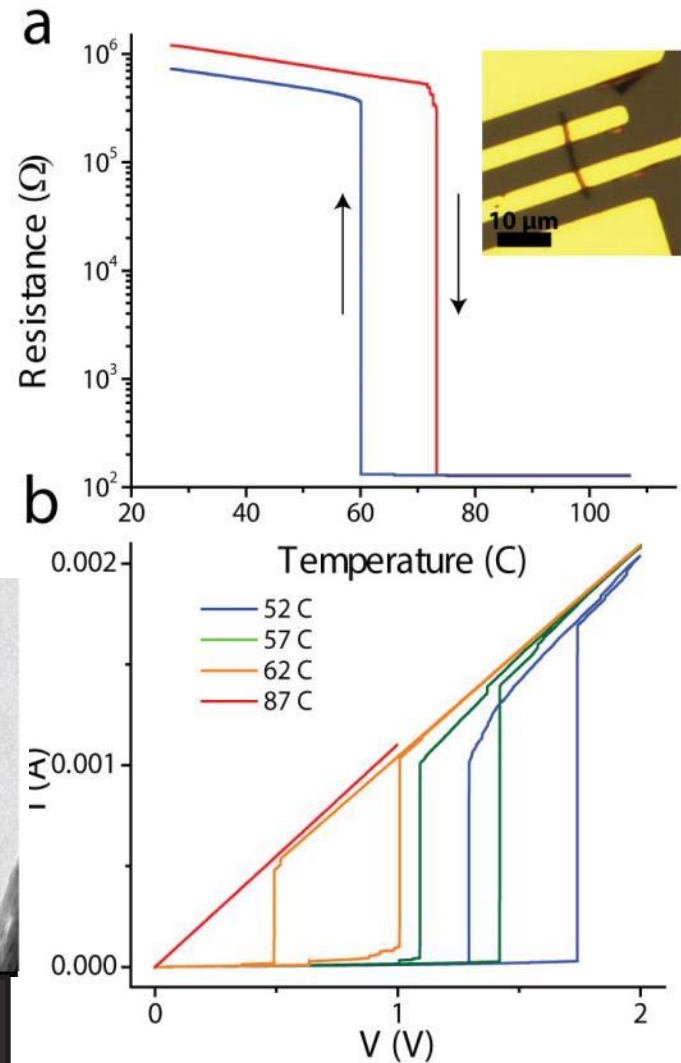
L. Whittaker, C. J. Patridge, and S. Banerjee, *J. Phys. Chem. Lett.* **2011**, *2*, 745-758

L. Whittaker, C. Jaye, Z. Fu, D. A. Fischer, and S. Banerjee, *J. Am. Chem. Soc.* **2009**, *131*, 8884-8894.

Phase Transitions in VO₂



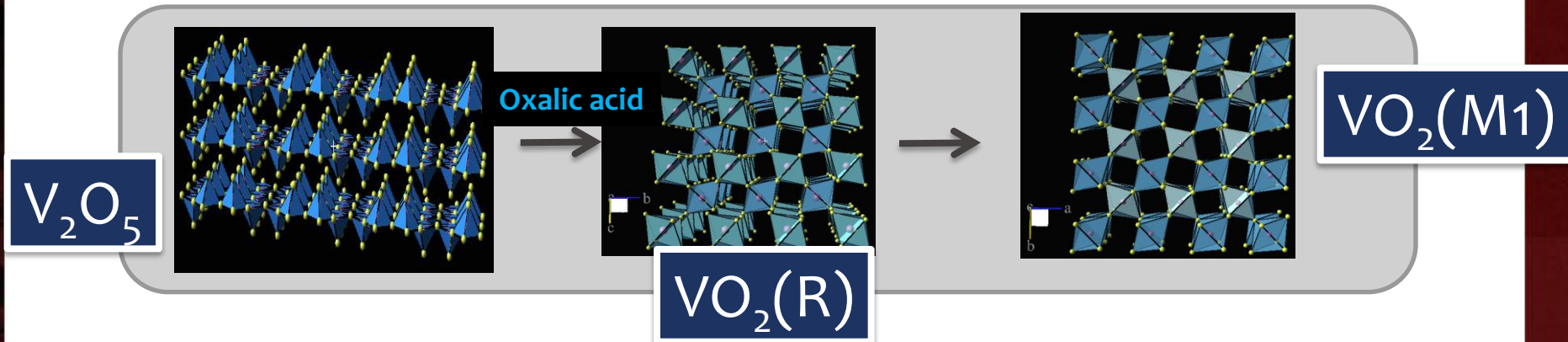
□ Almost four orders of magnitude
MIT: reproducible across different
nanowires and syntheses



G. A. Horrocks, S. K. Singh, M. F. Likely, G. Sambandamurthy, and S. Banerjee., *ACS Applied Materials and Interfaces*, 2014, 6, 15726–15732.

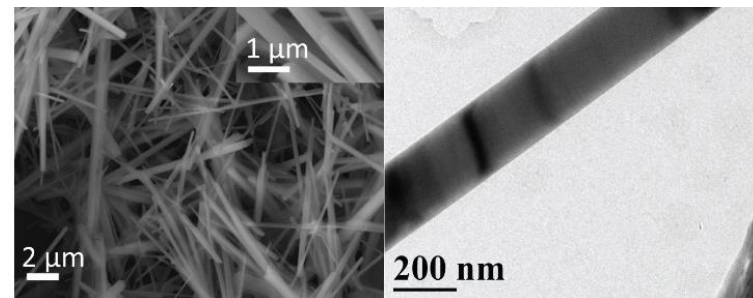
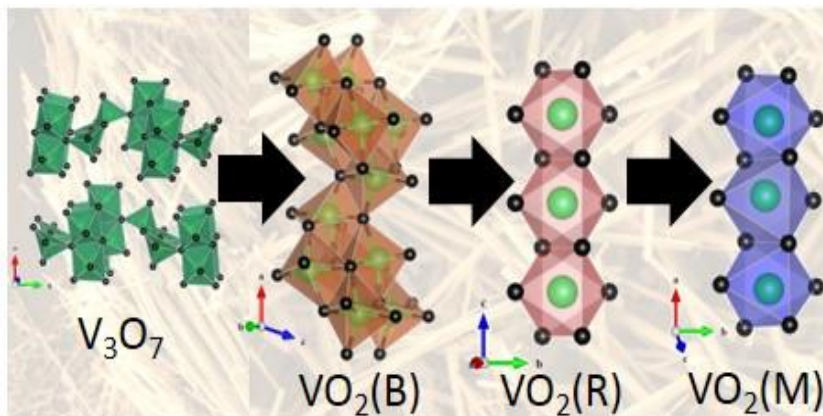


Synthesis & Scale-Up

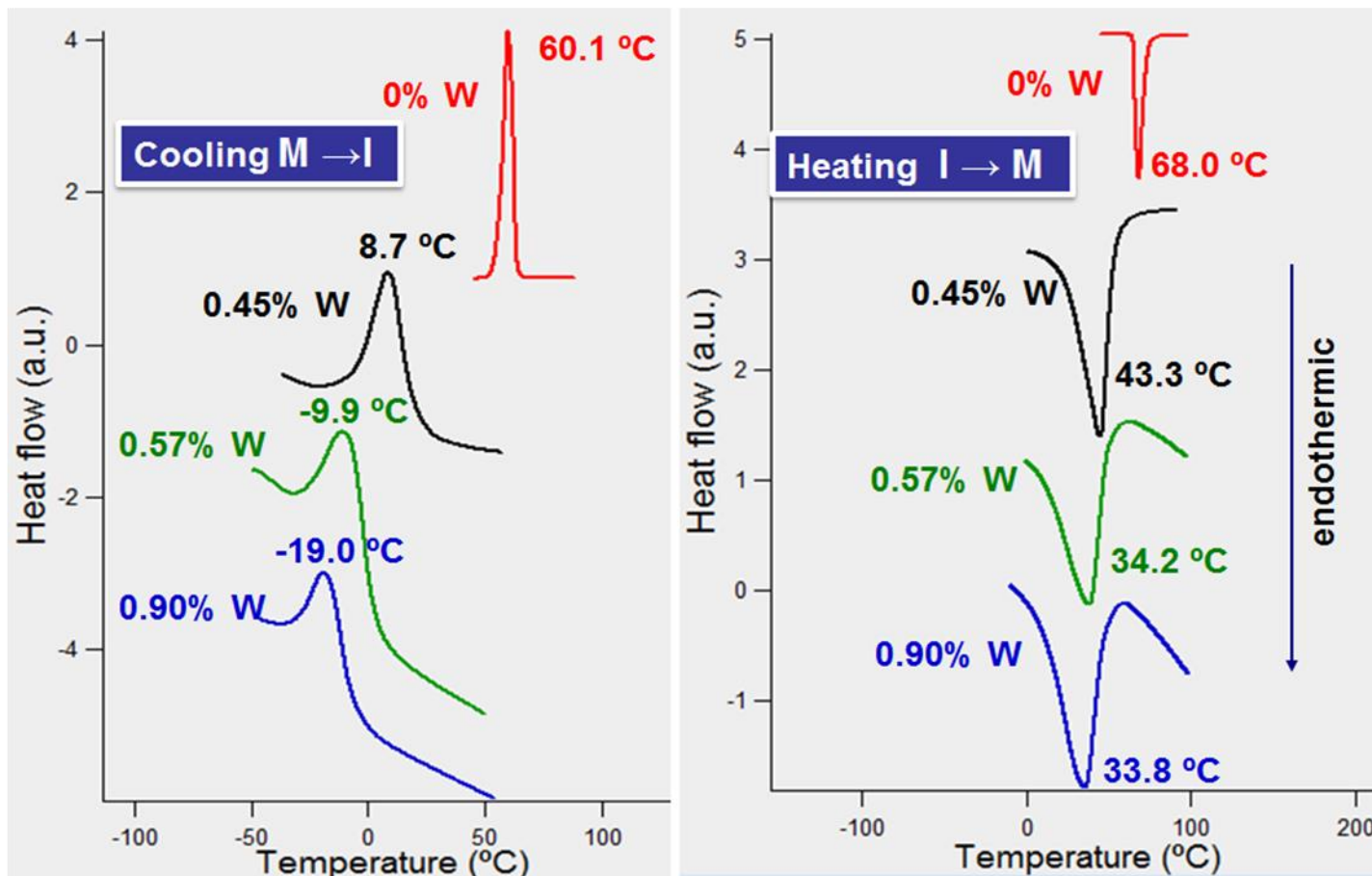


1. High-pressure route: One step, allows for introduction of W or Mo dopants

2. Low-pressure route: Two steps, can introduce dopants, high-quality materials



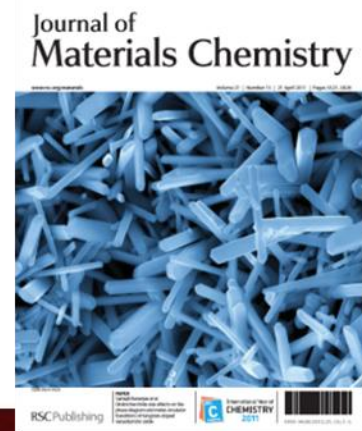
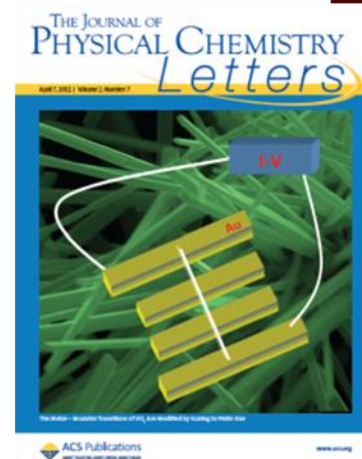
Depressing the Phase Transition



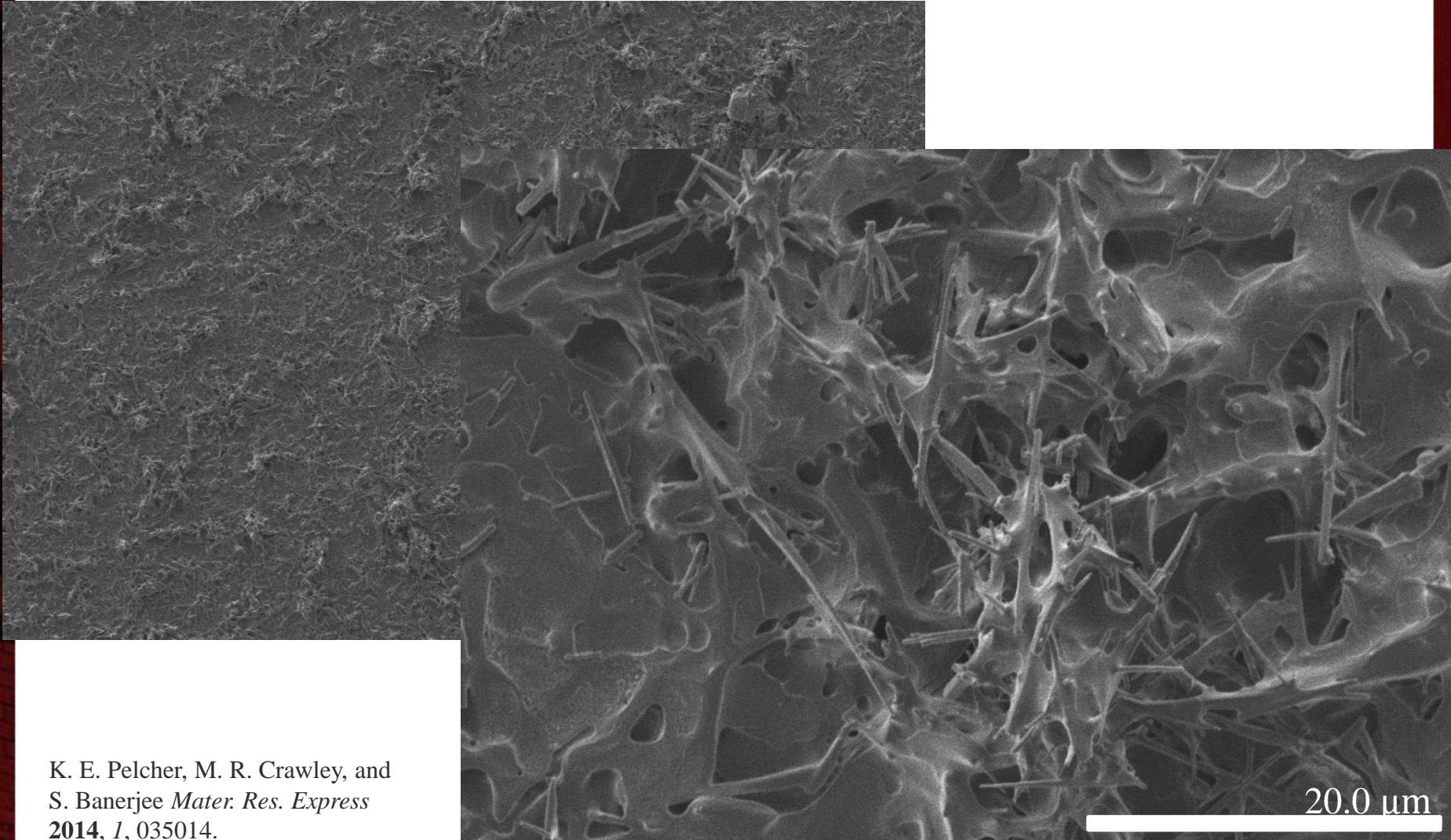
Customizable to different climates

L. Whittaker, T.-L. Wu, C. J. Patridge, G. Sambandamurthy, and S. Banerjee, *J. Mater. Chem.* **2011**, *21*, 5580.

L. Whittaker, T.-L. Wu, A. Stabile, G. Sambandamurthy, S. Banerjee, *ACS Nano* **2011**, 8861–8867

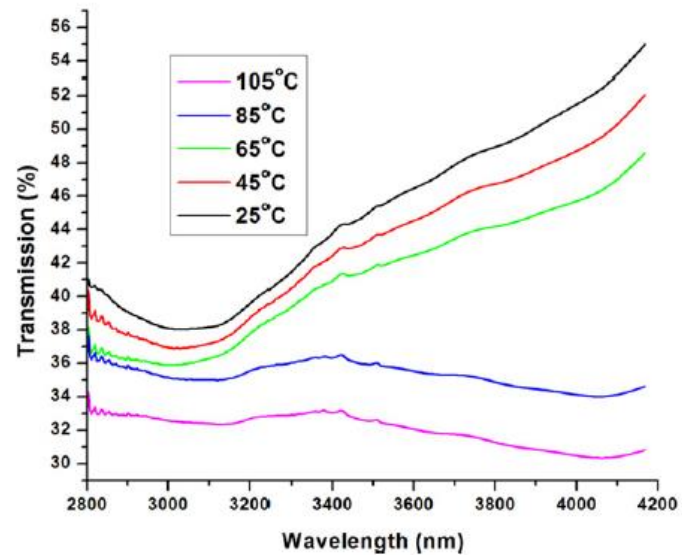
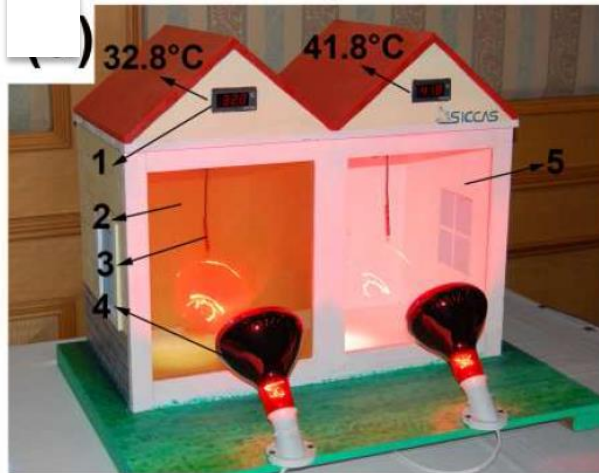


Adhering VO₂ to Glass: VO₂@SiO₂



K. E. Pelcher, M. R. Crawley, and
S. Banerjee *Mater. Res. Express*
2014, 1, 035014.

Optical Studies



Business

Panasci winners capitalize on high-tech coating

Rectangular Snip

Entrepreneurs take top prize in UB's Panasci competition

By Matt Glynn | News Business Reporter

on April 23, 2013 - 6:00 PM, updated April 23, 2013 at 11:50 PM

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Vayama
LONDON
BEST DEALS
SELECTED

The winning team in a University at Buffalo entrepreneurial competition pitched a plan to make a material coating that can regulate heat from the sun in a building, creating "smart" surfaces or windows.
Ann Brozek, Peter Marley and Brian Schultz won \$25,000 in start-up funding for their business, diMien LLC, in the Henry A. Panasci Jr. Technology Entrepreneurship Competition. A panel of judges selected the winners from five teams of finalists.

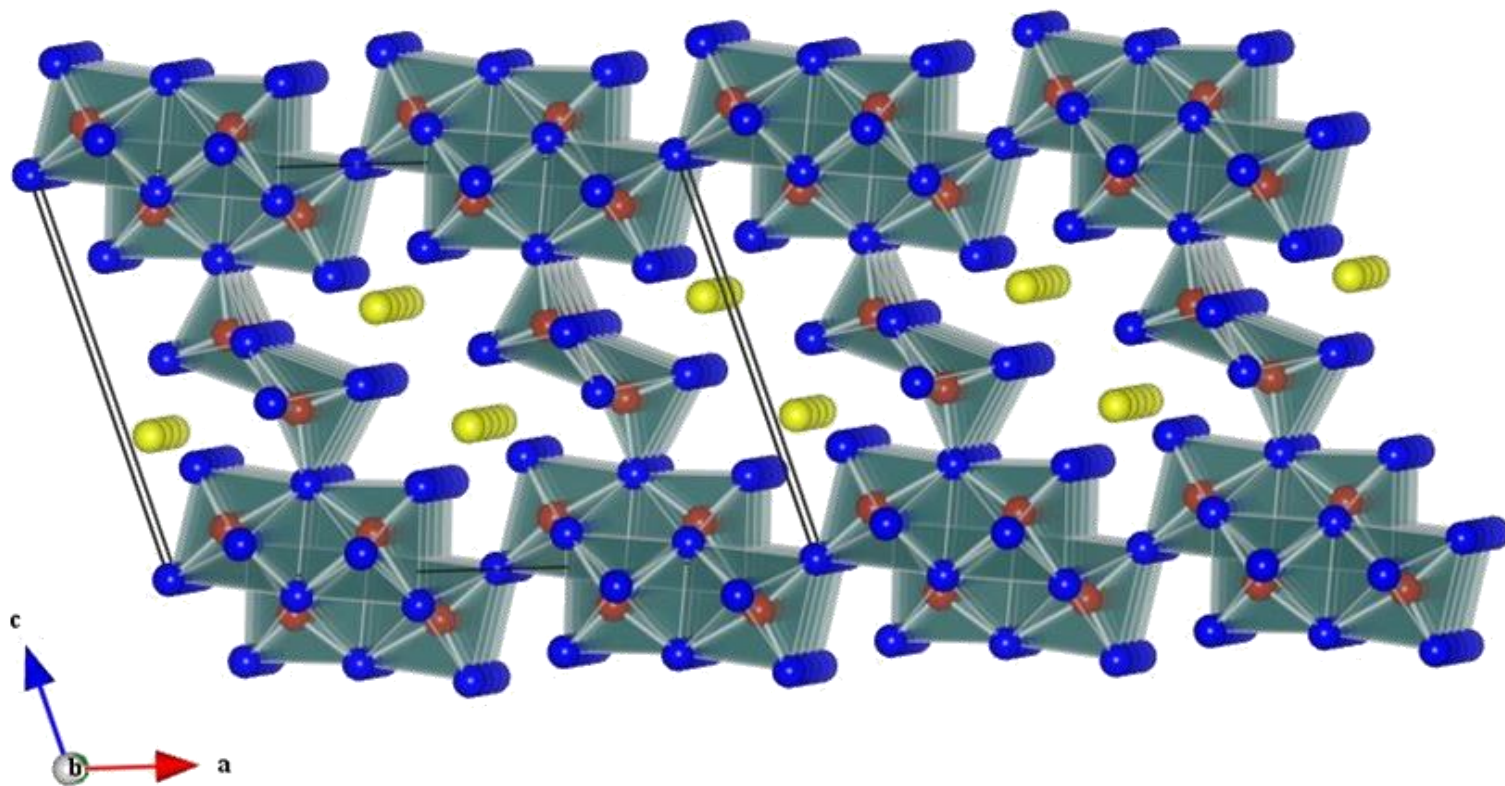


Peter Marley, Ph.D. '15
Ann Brozek, MBA. '14
Brian J. Schultz, Ph.D. '13

Alerts



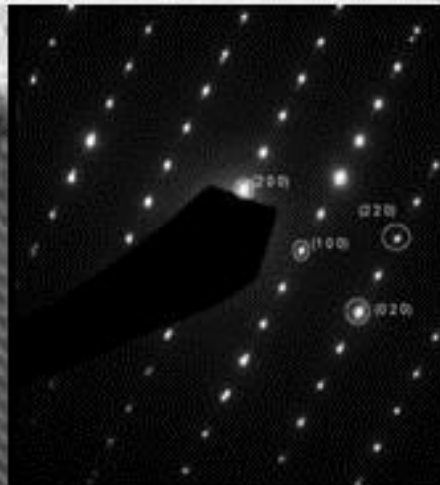
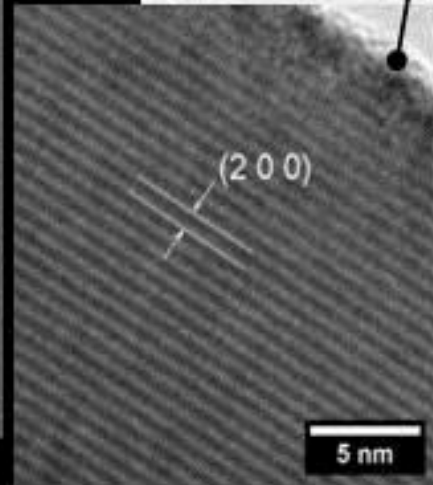
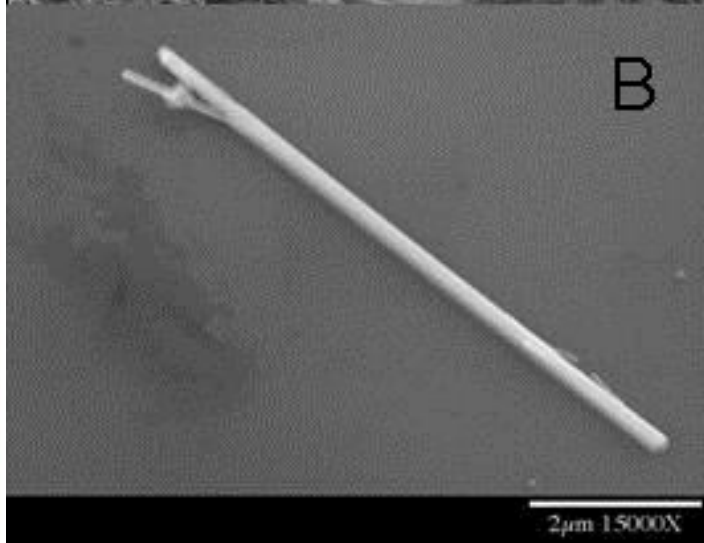
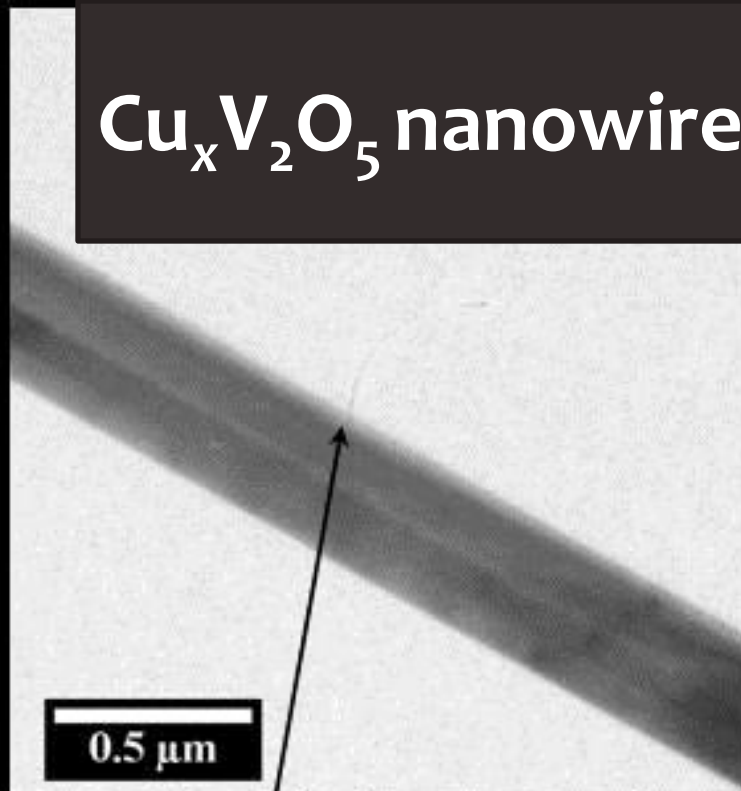
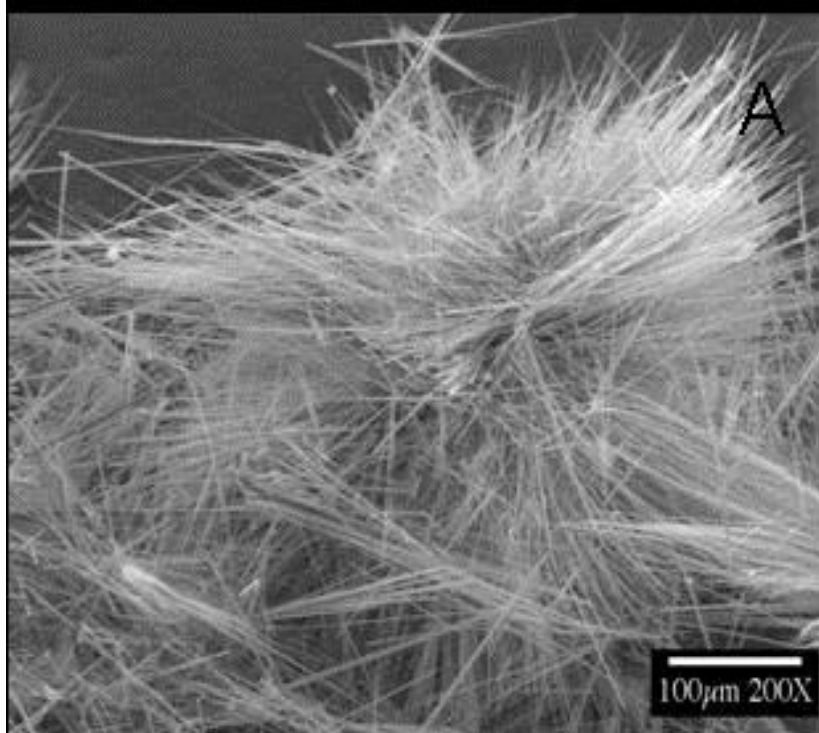
Life Beyond VO_2 : β' - $\text{Cu}_x\text{V}_2\text{O}_5$

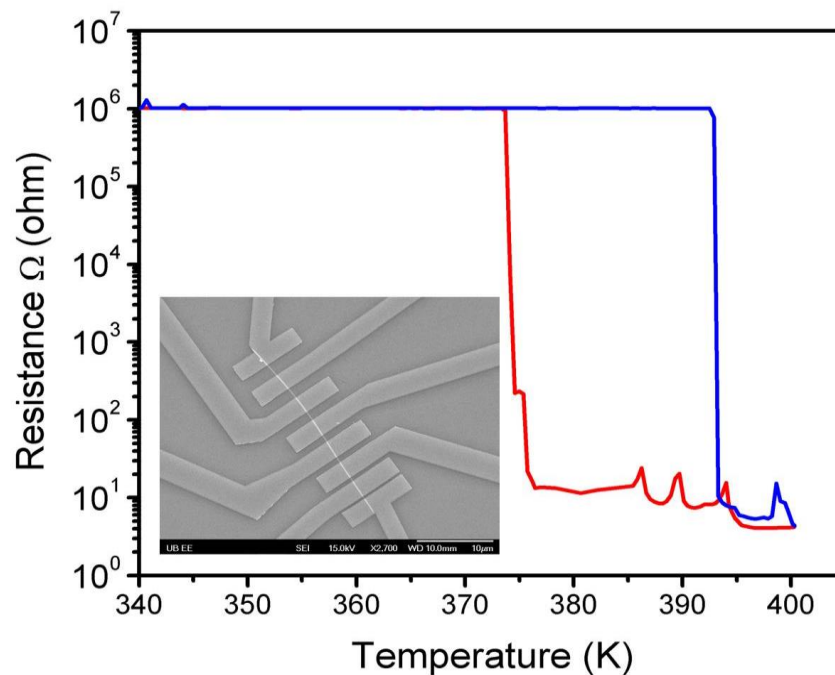
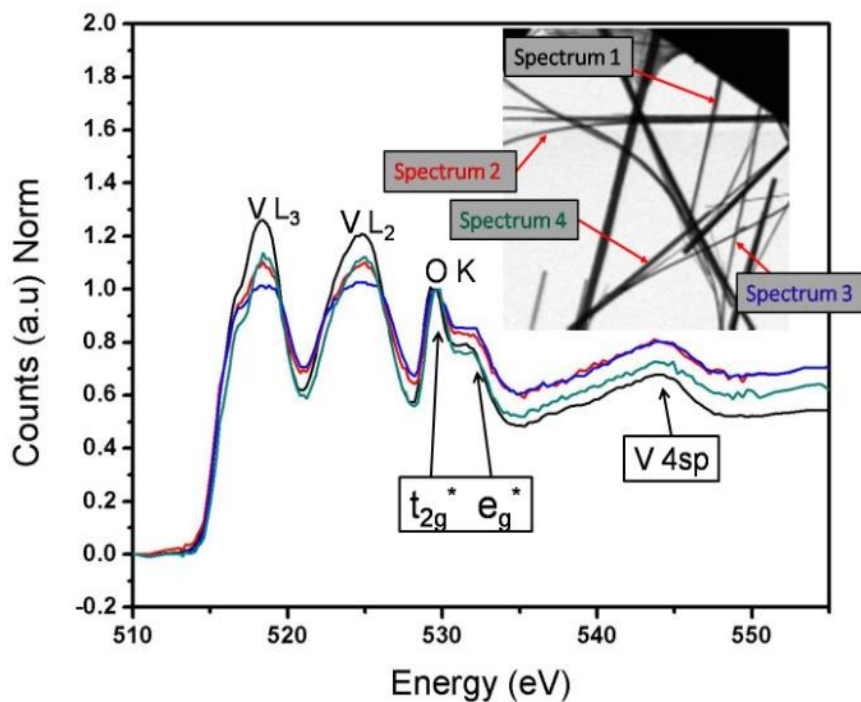


C. J. Patridge, C. Jaye, H. Zhang, A. C. Marschilok, D. A. Fischer, E. S. Takeuchi, and S. Banerjee, *Inorg. Chem.* 2009, 48, 3145-3152



$\text{Cu}_x\text{V}_2\text{O}_5$ nanowires





Colossal Metal–Insulator Transition in a Sienko-Wadsley Tunnel Bronze
 MIT Temperature and Magnitude Depend on Cu concentration
 in $\text{Cu}_x\text{V}_2\text{O}_5$

C. J. Patridge, C. Jaye, H. Zhang, A. C. Marschilok, D. A. Fischer, E. S. Takeuchi, and S. Banerjee, *Inorg. Chem.* **2009**, *48*, 3145-3152.

C. J. Patridge, T.-L. Wu, G. Sambandamurthy, and S. Banerjee, *Chem. Commun.* **2011**, *47*, 4484-4486.

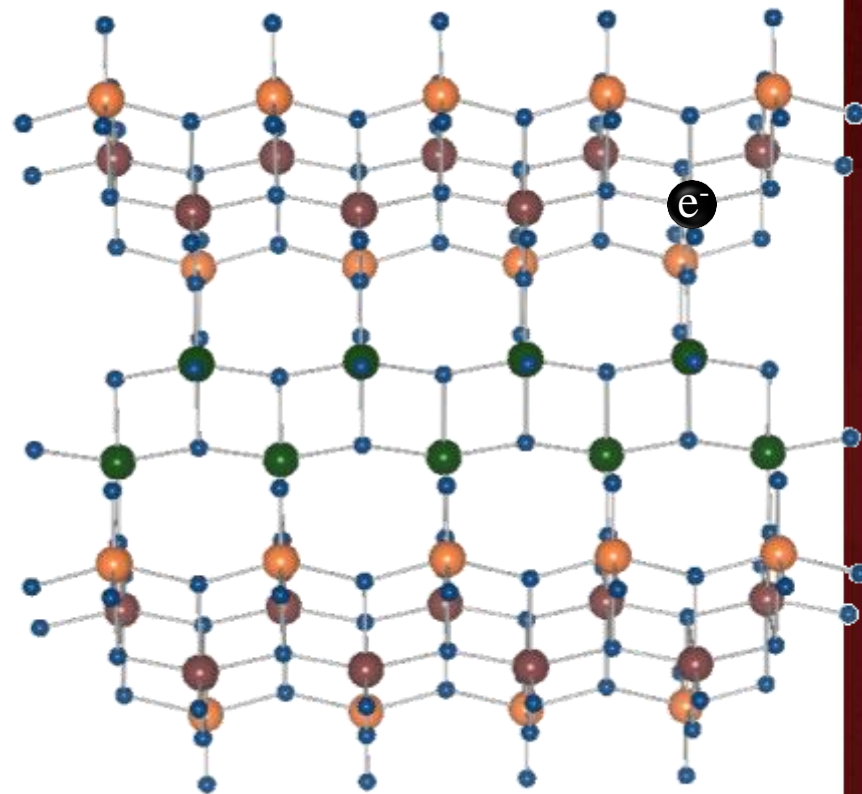


Some Mechanistic Ideas

In the insulating state localized V^{4+} (d^1 electrons) alternate with V^{5+} (d^0) cations along the vanadium oxide framework.

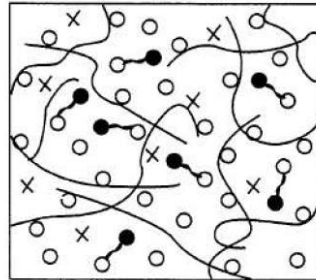
The specific mode of localization depends on the crystal structure and stoichiometry

Application of external stimuli results in reaching the Mott criterion and leads to a MIT.

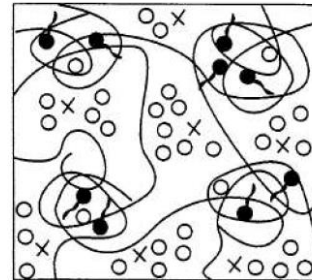


Thermotropic and Lyotropic Transitions

Water-clear



Paper-white



\rightleftharpoons
Phase transition

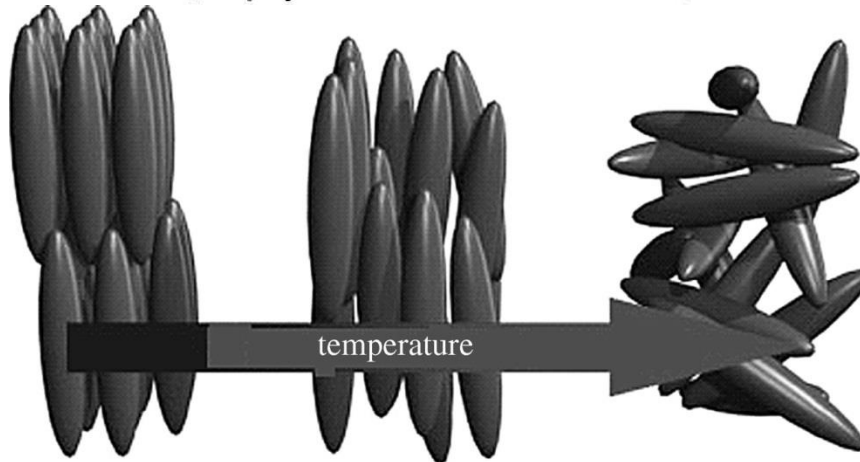
Isotropic aqueous solution
(Lower temperature)

Hydrogel with phase stability
(Higher temperature)

Phase segregation or change in assembly mode vastly alters refractive index

: Water : Water-soluble polymer

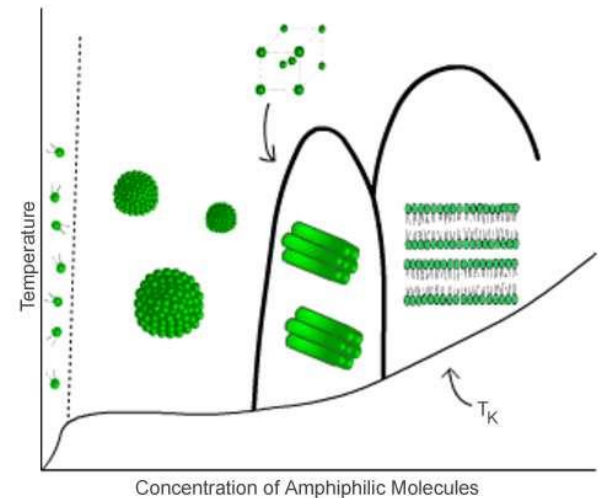
: Amphipathic molecule : NaCl



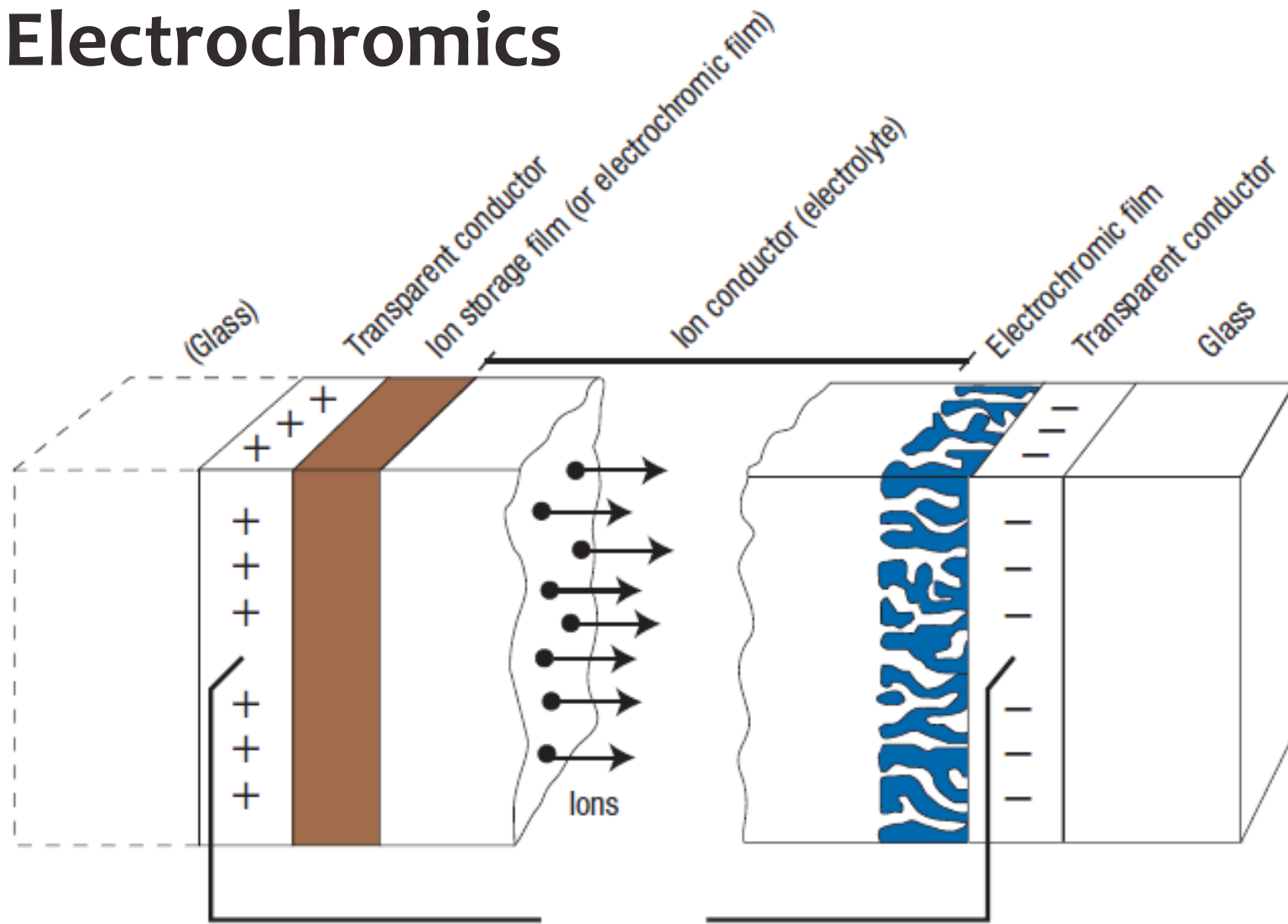
crystalline solid

liquid crystal

isotropic liquid



Electrochromics

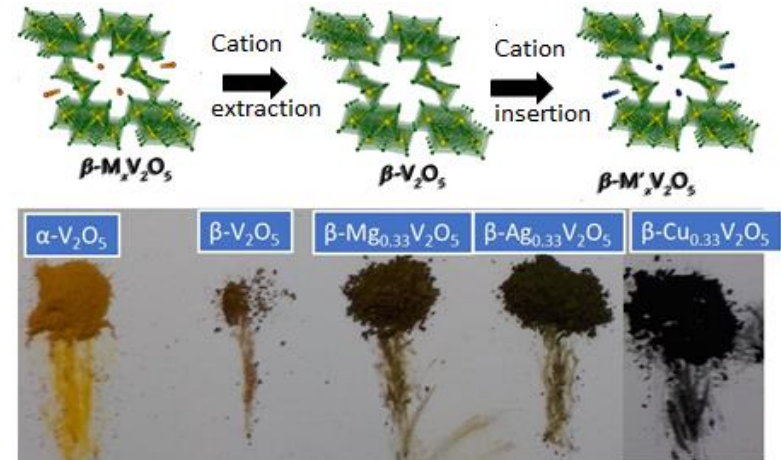


Granqvist, C. G.: Electrochromic Materials: Out of a Niche. *Nature Mater.* 2006, 5, 89-90.





Redox changes in active layers constituted of transition metal oxides: WO_3 , MoO_3 , V_2O_5

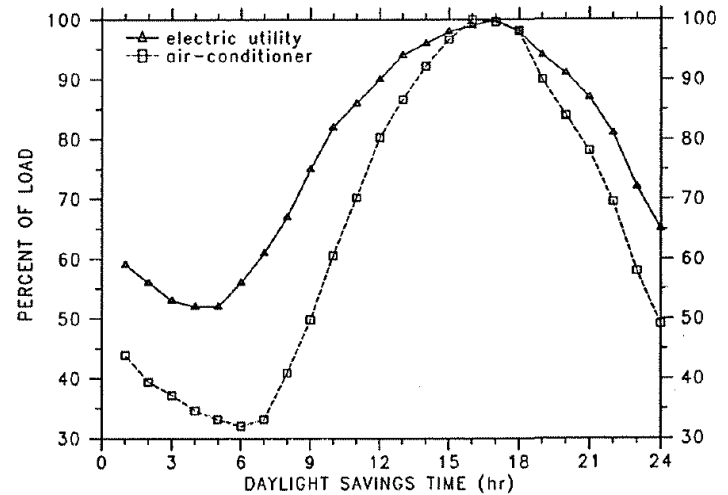
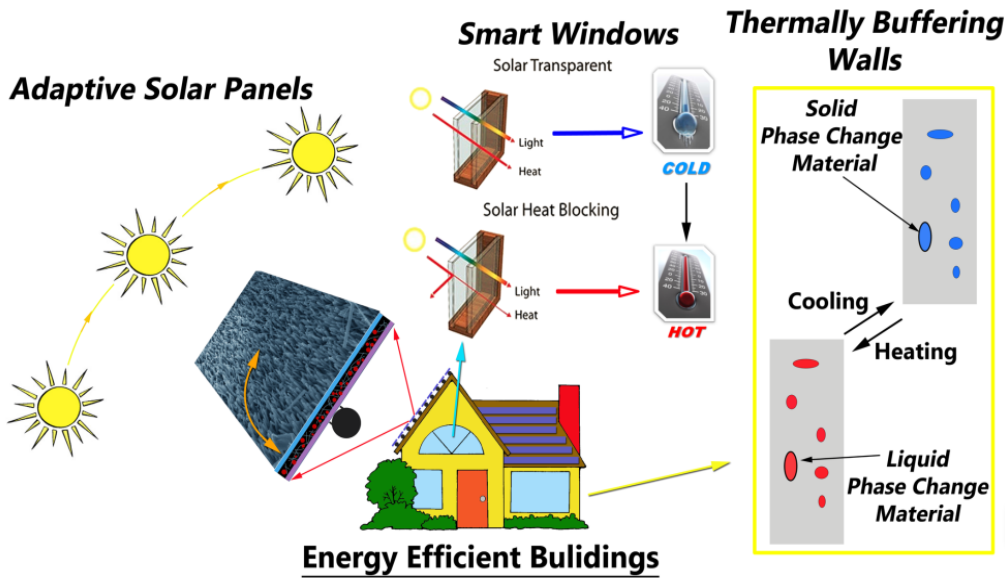


Not just windows...

Nairobi, Kenya



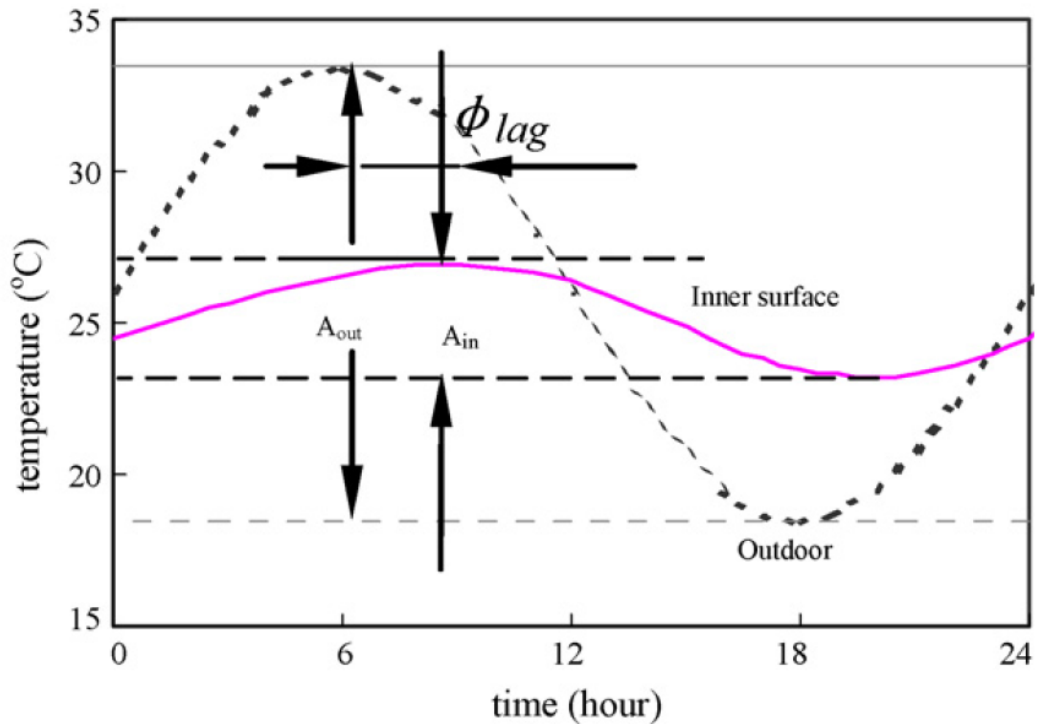
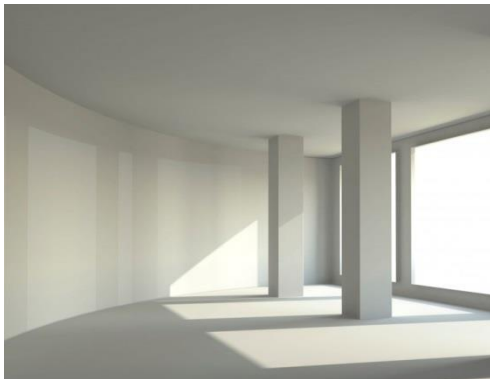
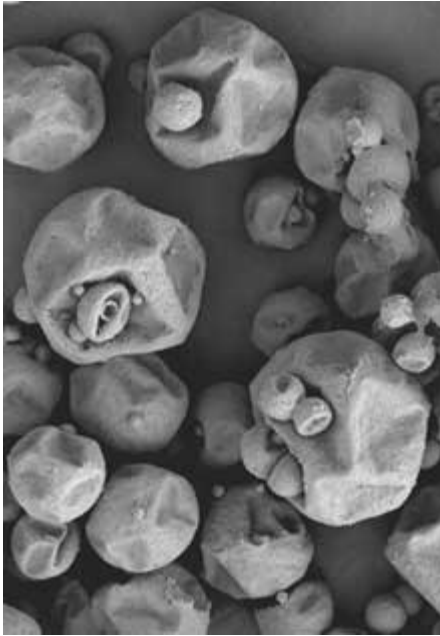
Buffering Heat Within Wallboards



Average peak-summer-day load shape for Florida electric utility and average load shape of 58 residential central air conditioners.

Melting of paraffins or salt hydrates allows for storage of thermal energy within walls

Phase Lag and Modulation



Energy and Buildings 40 (2008) 1771–1779

Image: Acrylic microcapsules from BASF

Phase change drywall from Universidad Politécnica de Madrid



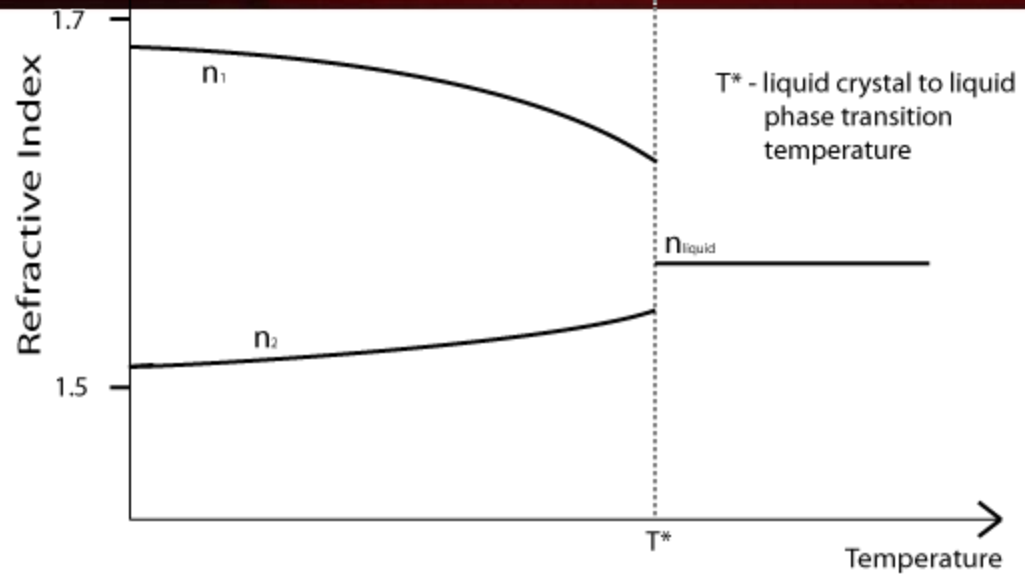
Conclusions

Designing responsive materials for building envelopes represents an underexplored opportunity: tremendous scope for phase-change materials (and for the scientists that like to play with them!)

Advances required in materials development, stability, building integration, and accurate stochastic modeling

Stakes are tremendously high from both energy efficiency and climate change perspective!





MEASURING THE TWO TYPES OF COATINGS

Low-E, %" airspace, %" clear	U-Value	VLT	SHGC	LSG
Pyrolytic	0.33 – 0.37	54% – 74%	0.45 – 0.66	1.09 – 1.25
Double-Silver MSVD (High VLT/Low Reflectance)	0.29 – 0.29	53% – 70%	0.28 – 0.39	1.76 – 1.98
Triple-Silver MSVD (High VLT/Low Reflectance)	0.28 – 0.29	61%	0.27 – 0.30	2.17 – 2.37